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Title: Treating acidemia in carbon monoxide poisoning

may be dangerous.

Authors: Peirce, EC

Keywords: Carbon Monoxide Poisoning/*therapy

Carbon Monoxide

CO

Hyperbaric Oxygenation

oxyhemoglobin dissociation curve

acid base temperature O2 availability

Issue Date: 1986

Publisher: Undersea and Hyperbaric Medical Society, Inc.

Citation: Peirce EC. Treating acidemia in carbon monoxide

poisoning may be dangerous. J Hyperbaric Med

. 1986; 1(2)87-97.

Abstract: In addition to reducing the available hemoglobin

(Hb) by tightly combining with it, CO poisoning

produces a substantial left shift in the

oxyhemoglobin dissociation curve. This reduces the usual ready O2 release in the tissue and is a major factor in the production of CNS hypoxia and damage. An academia, often present in severe CO poisoning especially if there are also burns, moderates the left curve shift and may, therefore, facilitate tissue oxygenation. Usual emergency room regimens routinely employ alkali in the treatment of academia, and there is a tendency to give large amounts of sodium bicarbonate to a patient in coma with a substantial academia. This is potentially a very dangerous practice in CO victims as an alkalemia, or even a reduction in an academia, will further shift the O2 dissociation curve and may aggravate and hypoxia, perhaps ensuring CNS damage. This paper gives the mathematical details of the curve shift produced by CO and pH change, and explains how to combine the two so that their total magnitude

Description:

known.

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and the effect on available tissue O2 can be

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