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变价态唐松草新碱一PVC膜离子选择电极的研制及解离常数的测定

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摘要:

本文研制了以四苯硼一唐松草新碱缔合物为电活性物质的变价态唐松草新碱—PVC膜电极。电极膜按电活性物 质:PVC: DBP为1:8:8组成。该电极在pH 5.0~6.0, I =0.05的NaCI—HCI溶液中Nernst响应范围为1×10⁻³~ 1×10⁻⁵ mol/L。电极斜率为58.2 mV/logc。检测限为2.5×10⁻⁶ mol/L。用直接电位法考察了TDH⁺,TDH₂CF⁺⁺共 存时溶液pH和电极斜率S的关系。用S—pH关系,测定了25℃, I=0.05时的Ka, 值为(2.5±0.2)×10⁻⁴,用 $\stackrel{2}{E-pH}$ 关系, 测定了25℃, I =0.05时的Ka₂值为(8.1±0.9)×10⁻⁸。

关键词: 唐松草新碱 变价态离子选择电极 解离常数

PREPARATION OF THE VARIABLE-VALENCY THALIDASINE(TD) PVC MEMBRANE SELECTIVE ELECTRODE AND ITS APPLICATION IN THE pka DETERMINATION OF THALI DASI NE

WL Ma; XO Gu and DS Su

Abstract:

The construction and performance characteristics of a new variable-valency TD-PVC selective electrode was described. It is based on the TPB. TD ion-pair complex, which was used as electro-active material. The electrode membrane was composed of the complex, PVC and DBP in the ratio of 1:8:8 (w/w/w). The 事 苏德森 electrode showed Nernstian response from 10⁻³ to 10⁻⁵ mol/L in NaCl—HC1 (I=0.05) solution over pH range of 5.0 to 6.0 with a slope of 58.2 mV/pC. The detection limit was found to be 2.5×10^{-5} mol/L. The electrode response was quite fast and stable, and the slopes of calibration curves were quite reproducible. It was observed that the E-pC relation can not be simply described by Nernstian equation over wide pH range. The slopes of electrode response and electrode potentials varied in concordance with pH value of the solution. By using the S-pH relation, the Ka_1 was found to be $(2.5\pm0.2)\times10^{-4}$. $(25^{\circ}\text{C}, I=0.05)$. Accordingly, Ka₂ was found to be $(8.1\pm0.9)\times10^{-8}(25^{\circ}\text{C}, I=0.05)$ by using E-pH relation. Keywords: Varible-valency ion-selective electrode Dissociation constant Thalidasine

收稿日期 1987-01-05 修回日期 网络版发布日期

DOI:

基金项目:

通讯作者:

作者简介:

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