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ABSTRACT Fly ash, obtained from a thermal power plant, Anpara, Sonebhadra, India has been used as an effective ad- sorbent for the removal of malathion from aqueous solutions. The time required to attain equilibrium was found to increase from 40 to 60 minutes as the initial malathion concentration increases from 1 to 10 mg/L. The optimum pH value for adsorption was 4.50. The removal of malathion increased by increasing the tem- perature indicating endothermic nature of removal process. The fly ash exhibited first order rate kinetics and followed both Langmuir and Freundlich isotherm models. Endothermic nature of adsorption process was						
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further	supported from increasing valu	es of Langmuir and	d Freundlich constants	with increase in		
temperature. The adsorbent can be used as an economical product for the removal of malathion from wastewater also. A comparison of the adsorption capacity of fly ash with other adsorbents shows that fly					Downloads:	402,245
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V. Sing and Wa doi: 10	gh, R. Singh, P. Tiwari, J. Singh, F. Ge aste Water Using Fly Ash," <i>Journal</i> 0.4236/jwarp.2010.24037.	ode and Y. Sharma, "R of Water Resource and	emoval of Malathion from <i>Protection</i> , Vol. 2 No. 4, 1	Aqueous Solutions 2010, pp. 322-330.		
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