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Dechlorination of Trichloroethylene in Groundwater by Nanoscale Bimetallic Fe/Pd Particles

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ABSTRACT

Palladium/iron bimetallic nanoparticles were synthesized using microemulsion method in the water-in-oil (W/O) microemulsion system, which was made up of iso-octane, cetyltrimethyl-ammonium bromide (CTAB), butanol and water and characterized by measuring the conductivity of the solution. Transmission electron microscope (TEM) and energy dispersive X-ray microanalysis (EDX) analysis showed that the average diameter of synthesized palladium/iron bimetallic nanoparticles was less than 80 nm, which was much smaller than the particles produced by the solution method. The palladium/iron bimetallic nanoscale particles produced in the laboratory showed better performance on dechlorinating TCE than the other materials. The nanoscale Fe/Pd particles exhibited high reactivity. When Pd content is 0.5%, the best TCE dechlorination efficiency is achieved within 30min. And Fe/Pd nanoparticles show persistent reaction activity in some sense.

KEYWORDS

Trichloroethylene, Microemulsion, Nanoscale Pd/Fe Particles

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References

- [1] [1] J. MunakataMarr, P. L. McCarty, and M. S. Shields, " Enhancement of trichloroethylene degradation in aquifer microcosms bioaugmented with wild type and genetically altered Burkholderia (Pseudomonas) cepacia G4 and PR1," *Environmental Science and Technology*, Vol. 30, No. 6, pp. 2045– 2052, 1996.
- [2] [2] A. K. Friis, H. J. Albrechtsen, and G. Heron, " Redox processes and release of organic matter after thermal treatment of a TCE-contaminated aquifer," *Environmental Science and Technology*, Vol. 39, No. 15, pp. 5787– 5795, 2005.
- [3] [3] H. Shen and J. T. Wilson, " Trichloroethylene removal from groundwater in flow-through columns simulating a permeable reactive barrier constructed with plant mulch," *Environmental Science and Technology*, Vol. 41, No. 11, pp. 4077– 4083, 2007.
- [4] [4] L. J. Matheson and P. G. Tratnyek, " Reductive dehalogenation of chlorinated methanes by iron metal," *Environmental Science and Technology*, Vol. 28, No. 12, pp. 2045– 2053, 1994.
- [5] [5] C. B. Wang and W. X. Zhang, " Synthesizing nanoscale iron particles for rapid and complete dechlorination of TCE and PCBs." *Environmental Science and Technology*, Vol. 31, No. 7, pp. 2154– 2156, 1997.
- [6] [6] T. L. Johnson, M. M. Scherer, and P. G. Tratnyek, " Kinetics of halogenated organic compound degradation by iron metal," *Environmental Science and Technology*, Vol. 30, No. 8, pp. 2634– 2640, 1996.

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- [7] [7] G. D. Sayles, G. You, M. Wang, and M. J. Kupferle, " DDT, DDD, and DDE dechlorination by zerovalent iron," *Environmental Science and Technology*, Vol. 31, No. 12, pp. 3448– 3454, 1997.
- [8] [8] W. A. Arnold and A. L. Roberts, " Pathways and kinetics of chlorinated ethylene and chlorinated acetylene reaction with Fe(0) particles," *Environmental Science and Technology*, Vol. 34, No. 9, pp. 1794– 1805, 2000.
- [9] [9] W. S. Orth and R. W. Gillham, " Dechlorination of tri-chloroethene in aqueous solution using Fe0," *Environmental Science and Technology*, Vol. 30, No. 1, pp. 66– 71, 1996.
- [10] [10] T. Li and J. Farrell, " Reductive dechlorination of tri-chloroethene and carbon tetrachloride using iron and palladium-iron cathodes," *Environmental Science and Technology*, Vol. 34, No. 1, pp. 173– 179, 2000.
- [11] [11] P. Zhang, X. Tao, Z. Li, and R. S. Bowman, " Enhanced perchloroethylene reduction in column systems using surfactant-modified zeolite/zero-valent iron pellets," *Environmental Science and Technology*, Vol. 36, No. 16, pp. 3597– 3603, 2002.
- [12] [12] G. V. Lowry and M. Reinhard, " Pd-catalyzed TCE dechlorination in groundwater: solute effects, biological control, and oxidative catalyst regeneration," *Environmental Science and Technology*, Vol. 34, No. 15, pp. 3217– 3223, 2000.
- [13] [13] W. X. Zhang, C. B. Wang, and H. L. Lien, " Treatment of chlorinated organic contaminants with nanoscale bimetallic particles," *Catalogue Today*, Vol. 40, No. 4, pp. 387– 395, 1998.
- [14] [14] Y. Xu and W. X. Zhang, " Subcolloidal Fe/Ag particles for reductive dehalogenation of chlorinated benzenes," *Industrial And Engineering Chemistry Research*, Vol. 39, No. 7, pp. 2238– 2244, 2000.
- [15] [15] B. Schrick, J. L. Blough, A. D. Jones, and T. E. Mallouk, " Hydrodechlorination of trichloroethylene to hydrocarbons using bimetallic nickel-iron nanoparticles," *Chemistry of Materials*, Vol. 14, No. 12,