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Abstract

The VdW-711 equation of state for polar compounds of **Androulakis** et al. (1989) is combined with the UNIQUAC model for the excess Gibbs free energy to establish the mixing rule in the attractive term of the equation of state. The obtained model, applicable thus to both vapour and liquid phases, eliminates the uncertainty in estimating the vapour phase non-ideality, a serious problem in using the gamma—phi approach, especially at higher pressures. Application of the model: (i) the correlation of binary vapour—liquid equilibria (VLE) data of polar systems and (ii) the prediction of multicomponent VLE behaviour from binary data, gives very good results, comparable to those with the gamma—phi approach, in both cases. In addition, successful prediction of saturated liquid volumes is obtained. Prediction of liquid volumes of mixing is also discussed.

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