

## Functional characterization of $\text{Cl}^-/\text{HCO}_3^-$ exchange in villous cells of the mouse ileum

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(Received 15 August 2006; and accepted 29 September 2006)

### ABSTRACT

At least three kinds of  $\text{Cl}^-/\text{HCO}_3^-$  exchangers, SLC26A3, SLC26A6 and AE2, have been demonstrated to be expressed in the intestinal epithelial cell. To examine the functional expression of these exchangers in the native enterocyte, we studied the  $\text{Cl}^-/\text{HCO}_3^-$  exchange activity in isolated villi from the mouse ileum by microfluorometric intracellular pH ( $\text{pH}_i$ ) measurement. The  $\text{pH}_i$  value increased upon  $\text{Cl}^-$  removal when the villus was superfused with an  $\text{HCO}_3^-/\text{CO}_2$ -buffered solution, while the response was blunted when superfused with an  $\text{HCO}_3^-/\text{CO}_2$ -free, Hepes-buffered solution. The recovery of  $\text{pH}_i$  value induced by  $\text{Cl}^-$  re-addition (after initial  $\text{Cl}^-$  removal) was totally or partially mimicked by the addition of  $\text{Br}^-$ ,  $\text{I}^-$ ,  $\text{F}^-$ ,  $\text{NO}_3^-$ , or  $\text{SO}_4^{2-}$  (in the absence of  $\text{Cl}^-$ ). The increase in  $\text{pH}_i$  value induced by  $\text{Cl}^-$  removal was partially inhibited in the presence of DIDS (30  $\mu\text{M}$ ), tenidap (10  $\mu\text{M}$ ), niflumic acid (30  $\mu\text{M}$ ) or NPPB (30  $\mu\text{M}$ ). Increasing the  $\text{K}^+$  concentration from 5 mM to 60 mM in the superfusion solution induced a reversible increase in  $\text{pH}_i$  value under the  $\text{HCO}_3^-/\text{CO}_2$ -buffered condition, while it had hardly any effect on  $\text{pH}_i$  under the Hepes-buffered condition. The  $\text{K}^+$ -induced  $\text{pH}_i$  changes were partially suppressed by removing  $\text{Cl}^-$  from the superfusion solution. These results, together with the reported findings of mouse *slc26a3*, *slc26a6* and AE2 in heterologously expressed systems, suggest the possibility that these three exchangers may all be functionally expressed in mouse ileal villous cells.

Villous enterocytes in the small intestine demonstrate plasma membrane  $\text{Cl}^-/\text{HCO}_3^-$  exchange activity which is believed to be involved in absorbing NaCl by coupling with  $\text{Na}^+/\text{H}^+$  exchange, in protecting mucosa by mediating  $\text{HCO}_3^-$  secretion and in regulating the intracellular pH value (5, 10, 21, 30, 31). Three kinds of membrane  $\text{Cl}^-/\text{HCO}_3^-$  exchangers have recently been identified in the mammalian small intestine, *i.e.*, SLC26A6 (alias PAT1), SLC26A3 (alias DRA) and AE2 (alias SLC4A2) (11, 13, 15, 17, 20, 22, 23, 26, 28, 33). Both SLC26A6

and SLC26A3 are mainly expressed in the apical membrane, and are assumed to be responsible for  $\text{Cl}^-$  absorption and  $\text{HCO}_3^-$  secretion (11, 32). On the other hand, AE2 is expressed mainly in the basolateral membrane and probably exercises the house-keeping function of intracellular pH regulation (1, 2). The transport and pharmacological characteristics of these three  $\text{Cl}^-/\text{HCO}_3^-$  exchangers have been partially revealed by using the heterologously expressed systems (6, 7, 9, 12, 14, 17, 18, 27, 34).

The objective of this present study is to examine whether SLC26A6, SLC26A3 and AE2 actually operate in the native enterocyte. To do this, we evaluated the  $\text{Cl}^-/\text{HCO}_3^-$  exchange activity in isolated mouse villi by microfluorometric intracellular pH ( $\text{pH}_i$ ) measurement (8). We examined the anion selectivity, inhibitor sensitivity, and electrogenicity of  $\text{Cl}^-/\text{HCO}_3^-$  exchange to compare with the reported

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observations for mouse *slc26a6*, *slc26a3*, and *AE2* by using heterologously expressed systems. We specifically selected the ileum, since this segment has been relatively disregarded in comparison with the proximal small intestine (11, 29, 32, 33) and the proximal and distal small intestinal actions may not be identical.

## MATERIALS AND METHODS

**Tissue preparation.** The present experiments were performed under the Guiding Principles for the Care and Use of Animals in the Field of Physiological Sciences recommended by the Physiological Society of Japan. Male mice (*ddy*; Japan SLC, Shizuoka, Japan) weighing 35–40 g were acclimatized on a standard diet, with food and water being provided *ad libitum*. The animals were killed by dislocation of the cervical vertebrae. A segment of the distal ileum was excised, opened and kept in an oxygenated HEPES-buffered solution. Several tips of villi were cut from each ileal segment by microdissection scissors under a stereomicroscope while the intestinal segment was pinned with the mucosal side upward, as described previously for the guinea-pig ileum (8). These were subsequently used for measuring the  $pH_i$  value and for an RT-PCR analysis.

**RT-PCR analysis.** Total RNA was extracted from the villus tips by using an RNeasy<sup>®</sup> micro kit (Qiagen, Tokyo, Japan) according to the manufacturer's instructions. The RNA samples were then subjected to the RT-PCR reaction with a OneStep RT-PCR kit (Qiagen) according to the manufacturer's instructions. The primers used in the RT-PCR analysis were as follows: mouse *slc26a6*, forward primer 5'-TGTACTTCGCCAATGCTGAGC-3' and reverse primer 5'-GTCCAGGATGAGGTATGGA-3' (Genbank Accession No. NM134420); mouse *slc26a3*, forward primer 5'-TTCCTGGACCATCTCAAAGG-3' and reverse primer 5'-TGTAACGACAACTCCCACCA-3' (Genbank Accession No. NM021353); mouse *AE2*, forward primer 5'-ATTGAGGAAGGGGAGGAAGA-3' and reverse primer 5'-GTCTTCGCTCCTGAAGGTTG-3' (Genbank Accession No. NM009207); mouse *GAPDH*; forward primer 5'-GTGTTCTACCCCAATGTG-3' and reverse primer 5'-TGTGAGGGAGATGCTCAGTG-3' (Genbank Accession No. NM008084). The amplification conditions were as follows: reverse transcription for 30 min at 50°C was followed by the initial PCR activation step for 15 min at 95°C, denaturation for 45 s at 94°C, annealing for 45 s at

55°C, and extension for 1 min at 72°C; 35 PCR cycles were performed. The PCR products were separated in 2% agarose gel with a Tris/acetic acid/EDTA buffer, and then detected by ethidium bromide.

**Measurement of  $pH_i$ .** The villus tip preparation was washed several times by centrifugation, before being re-suspended for 10 min in the dark at room temperature (20–24°C) in a HEPES-buffered solution containing 10  $\mu$ M of 2',7'-bis(2-carboxyethyl)-5(6)-carboxy-fluorescein acetoxymethyl ester (BCECF-AM) which had first been dissolved to a concentration of 10 mM in dimethyl sulfoxide. After loading the dye, the villus tip preparation was washed with the HEPES-buffered solution and subjected to intracellular pH measurement (8). The suspension of dye-loaded villi was first seeded on a specially designed perfusion vessel, which had been coated with a polylysine cell adhesive (Poly-L-Lysine Solution, Sigma), before the vessel was centrifuged at 200 g for 5 min at room temperature to fix the villus preparation. The vessel was then placed on the stage of an inverted microscope (TMD, Nikon, Tokyo, Japan) equipped with a CAM-230 dual-wavelength fluorometric system (Japan Spectroscopic, Tokyo, Japan), before being perfused at 6 mL/min with a gas-equilibrated solution and maintained at 34°C by pre-warming the inflow line. The whole area of a selected villus tip was alternately illuminated at 440 and 500 nm for 50 ms by a chopper at 1000 Hz, and the fluorescence was measured at 520–570 nm through a band-pass filter. The  $pH_i$  value was calculated from the mean fluorescence ratio (F500/F440) every 2 s. This complete procedure was controlled by a computer (Macintosh LC) which was equipped with a Lab View 2 data acquisition and analysis system (National Instruments, Houston, TX, USA). There was negligible autofluorescence from the villus tip.

The  $pH_i$  level was calibrated *in situ* by the nigericin- $K^+$  method. Dye-loaded villus tips were incubated with nigericin (20  $\mu$ M) for 5 min, and then with a high- $K^+$  perfusion solution of various pH values (KCl, 130 mM; NaCl, 10 mM;  $CaCl_2$ , 1.5 mM; Mg  $Cl_2$ , 1.0 mM; HEPES or MES, 10 mM; pH was adjusted with Tris at 37°C) to determine the relationship between the fluorescence ratio (F500/F440) and  $pH_i$  value. Calibration curves were obtained over a pH range of 6–8.

The  $Cl^-/HCO_3^-$ -exchange activity was determined from the  $pH_i$  change induced by removing  $Cl^-$  from the perfusing solution. We performed the  $Cl^-$ -remov-

al challenge two or three times on the same tissue, but under different experimental conditions. The results were analyzed after assuming that the time-dependent change in the response to Cl<sup>-</sup> removal was negligible. This assumption was confirmed to be valid by the response not being significantly different when the first and third challenges were performed under control conditions (*cf.* Figs. 3 and 6).

**Solutions and chemicals.** The HEPES-buffered standard solution had the following composition (mM): NaCl, 110; Na-gluconate, 25; KCl, 5; CaCl<sub>2</sub>, 1; MgCl<sub>2</sub>, 1; glucose 10; HEPES, 15; the pH value adjusted to 7.4 with NaOH and the solution was gassed with 100% O<sub>2</sub>. The CO<sub>2</sub>/HCO<sub>3</sub><sup>-</sup>-buffered standard solution had the following composition (mM): NaCl, 110; NaHCO<sub>3</sub>, 25; KCl, 5; CaCl<sub>2</sub>, 1; MgCl<sub>2</sub>, 1; glucose, 10; HEPES, 15; the solution was gassed with 95% O<sub>2</sub>-5% CO<sub>2</sub>. The Cl<sup>-</sup>-free solution was prepared by equimolar substitution of Cl<sup>-</sup> with gluconate and by increasing Ca-(gluconate)<sub>2</sub> concentration to 5 mM. The high-K<sup>+</sup> solution had the following composition (mM): NaCl, 55; NaHCO<sub>3</sub>, 25; KCl, 60; CaCl<sub>2</sub>, 1; MgCl<sub>2</sub>, 1; glucose, 10; HEPES, 15. The normal K<sup>+</sup> solution for the high-K<sup>+</sup> experiment had the following composition (mM): NaCl, 55; NaHCO<sub>3</sub>, 25; KCl, 5; CaCl<sub>2</sub>, 1; MgCl<sub>2</sub>, 1; glucose, 10; HEPES, 15; N-methyl-D-glucamine-Cl, 55. The normal K<sup>+</sup>, Cl<sup>-</sup>-free solution had the following composition (mM): Na-gluconate, 55; NaHCO<sub>3</sub>, 25; K-gluconate, 5; Ca-(gluconate)<sub>2</sub>, 5; Mg-(gluconate)<sub>2</sub>, 1; glucose, 10; HEPES, 15; mannitol, 110. The high K<sup>+</sup>, Cl<sup>-</sup>-free solution had the following composition (mM): Na-gluconate, 55; NaHCO<sub>3</sub>, 25; K-gluconate, 60; Ca-(gluconate)<sub>2</sub>, 5; Mg-(gluconate)<sub>2</sub>, 1; glucose, 10; HEPES, 15. The pH value of each of these solutions was adjusted to 7.4.

We obtained nigericin, 4,4'-diisothiocyanostilbene-2,2'-disulphonic acid (DIDS), and niflumic acid from Sigma (St. Louis, MO, USA), and BCECF-AM from Molecular Probes (Eugene, OR, USA). 5-Nitro-2-(3-phenylpropylamino)benzoic acid (NPPB) was obtained from RBI (Natick, MA, USA), and tenidap was generously presented by Pfizer (Groton, CT, USA).

**Statistical analysis.** Each experimental value is given as the mean ± SE (n = no. of animals). Two-group comparisons were analysed by a paired Student's two-tailed *t*-test. A probability value of < 0.05 has been considered significant in all instances.

## RESULTS

### *Expression of Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchangers in mouse ileal villi*

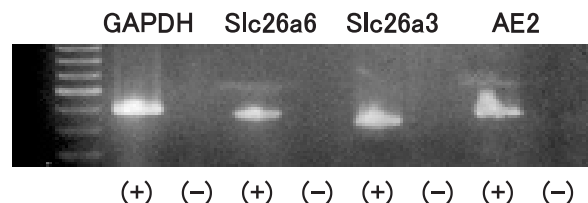
We first performed RT-PCR to detect the expression of mouse *slc26a6*, *slc26a3* and *AE2* mRNAs in the mouse ileal villus preparation which was used later for the pH<sub>i</sub> measurements. The primers were designed to obtain the expected PCR products of 371 bp, 324 bp and 349 bp for *slc26a6*, *slc26a3* and *AE2*, respectively. As shown in Fig. 1, the expected-size PCR fragment was amplified for each exchanger, demonstrating that mRNAs of *slc26a6*, *slc26a3* and *AE2* had all been expressed in the ileal villi.

### *Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity*

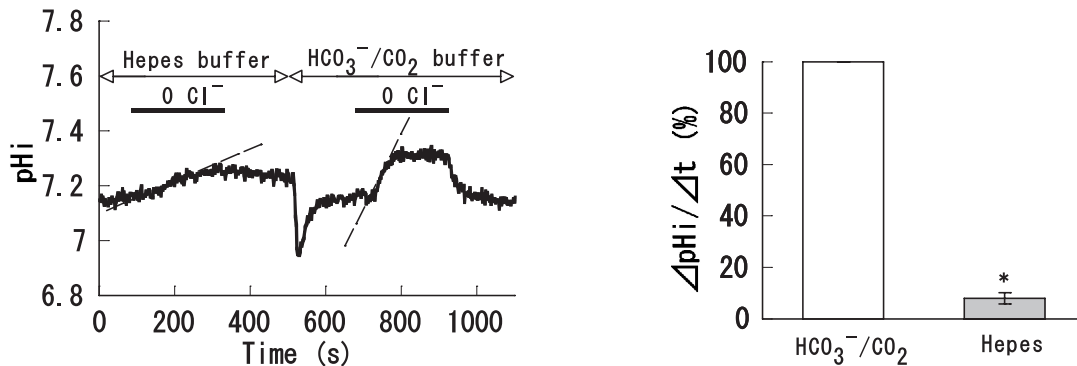
We then examined the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity in the ileal villous cells by microfluorometric pH<sub>i</sub> measurement, using the pH-sensitive dye BCECF. As shown in Fig. 2, the pH<sub>i</sub> value was rapidly increased by Cl<sup>-</sup> removal ( $\Delta\text{pH}_i/\text{min} = 0.24 \pm 0.03$ , n = 5) and then restored to the baseline level by Cl<sup>-</sup> re-addition ( $\Delta\text{pH}_i/\text{min} = -0.27 \pm 0.03$ ) when a villus was superfused with the HCO<sub>3</sub><sup>-</sup>/CO<sub>2</sub>-buffered solution. The magnitude of the increase in pH<sub>i</sub> was  $0.13 \pm 0.02$ . In contrast, the pH<sub>i</sub> increase upon Cl<sup>-</sup> removal was greatly diminished when superfused with the HCO<sub>3</sub><sup>-</sup>/CO<sub>2</sub>-free, HEPES-buffered solution. This indicates that the ileal villous cells had Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity, but little Cl<sup>-</sup>/OH<sup>-</sup>-exchange activity. The baseline pH<sub>i</sub> value was  $7.05 \pm 0.05$  (n = 5) in the HCO<sub>3</sub><sup>-</sup>/CO<sub>2</sub>-buffered solution and  $7.21 \pm 0.02$  (n = 5) in the HEPES-buffered solution, the former being significantly lower than the latter.

### *Anion selectivity of the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchanger*

We next examined the other anions, *i.e.*, Br<sup>-</sup>, Γ<sup>-</sup>, F<sup>-</sup>,



**Fig. 1** RT-PCR analysis of Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchanger mRNA expressed in the mouse ileal villus tips. The amplification products were size-fractionated on agarose gel and stained with ethidium bromide. The sizes of the expected products are as follows (in bp): GAPDH, 397; mouse *slc26a6*; 371; mouse *slc26a3*, 324; mouse *AE2*, 349. (+) and (-) respectively show the results with and without the RT treatment. The standard (left column) is marked in 100 bp with the thickest one being 500 bp.



**Fig. 2**  $pH_i$  changes induced by  $Cl^-$  removal in the presence and absence of  $HCO_3^-/CO_2$  in the superfusion solution. Left panel: A typical recording. The  $pH_i$  change for a single villus loaded with BCECF-AM was recorded. Gluconate was substituted for  $Cl^-$  in the  $Cl^-$ -free solution. Right panel: the rates of increase in  $pH_i$  ( $\Delta pH_i/\Delta t$ ) upon  $Cl^-$  removal were compared in the  $HCO_3^-/CO_2$ -buffered solution and in the  $HCO_3^-/CO_2$ -free, Hepes-buffered solution. \* $p < 0.05$ ,  $n = 5$ .

$NO_3^-$ , and  $SO_4^{2-}$ , for their ability to induce  $HCO_3^-$  efflux. As shown in Figs. 3A-E, all these anions caused a  $pH_i$  decrease when added in the absence of  $Cl^-$ , indicating that, like  $Cl^-$  they could activate  $HCO_3^-$  efflux. The rate of  $pH_i$  decrease induced by each of these anions was significantly lower than that induced by  $Cl^-$  (Fig. 3F). The steady  $pH_i$  level after adding  $F^-$  or  $NO_3^-$  was not significantly different from the baseline  $pH_i$ ; subsequent switching to the  $Cl^-$ -containing solution did not affect the  $pH_i$  value (Figs. 3C, D and G). On the other hand, the steady  $pH_i$  level after adding  $Br^-$ ,  $I^-$ , or  $SO_4^{2-}$  was significantly higher than the baseline level. Subsequent switching to the  $Cl^-$ -containing solution further decreased  $pH_i$  to a level not different from the baseline (Figs. 3A, B, E and G).

We then tried to confirm the presence of  $SO_4^{2-}/HCO_3^-$ -exchange activity, since mouse *slc26a6* has been shown to transport  $SO_4^{2-}$  (7, 12, 35). In this experiment,  $Cl^-$  was absent throughout the measurement (Fig. 4). The  $pH_i$  value was increased when  $SO_4^{2-}$  was removed and restored to the original level when  $SO_4^{2-}$  was re-administered. The response upon  $SO_4^{2-}$  removal was more prominent in the  $HCO_3^-/CO_2$ -buffered solution than in the  $HCO_3^-/CO_2$ -free, Hepes-buffered solution, confirming the presence of  $SO_4^{2-}/HCO_3^-$ -exchange activity.

#### Effects of inhibitors on $Cl^-/HCO_3^-$ exchange

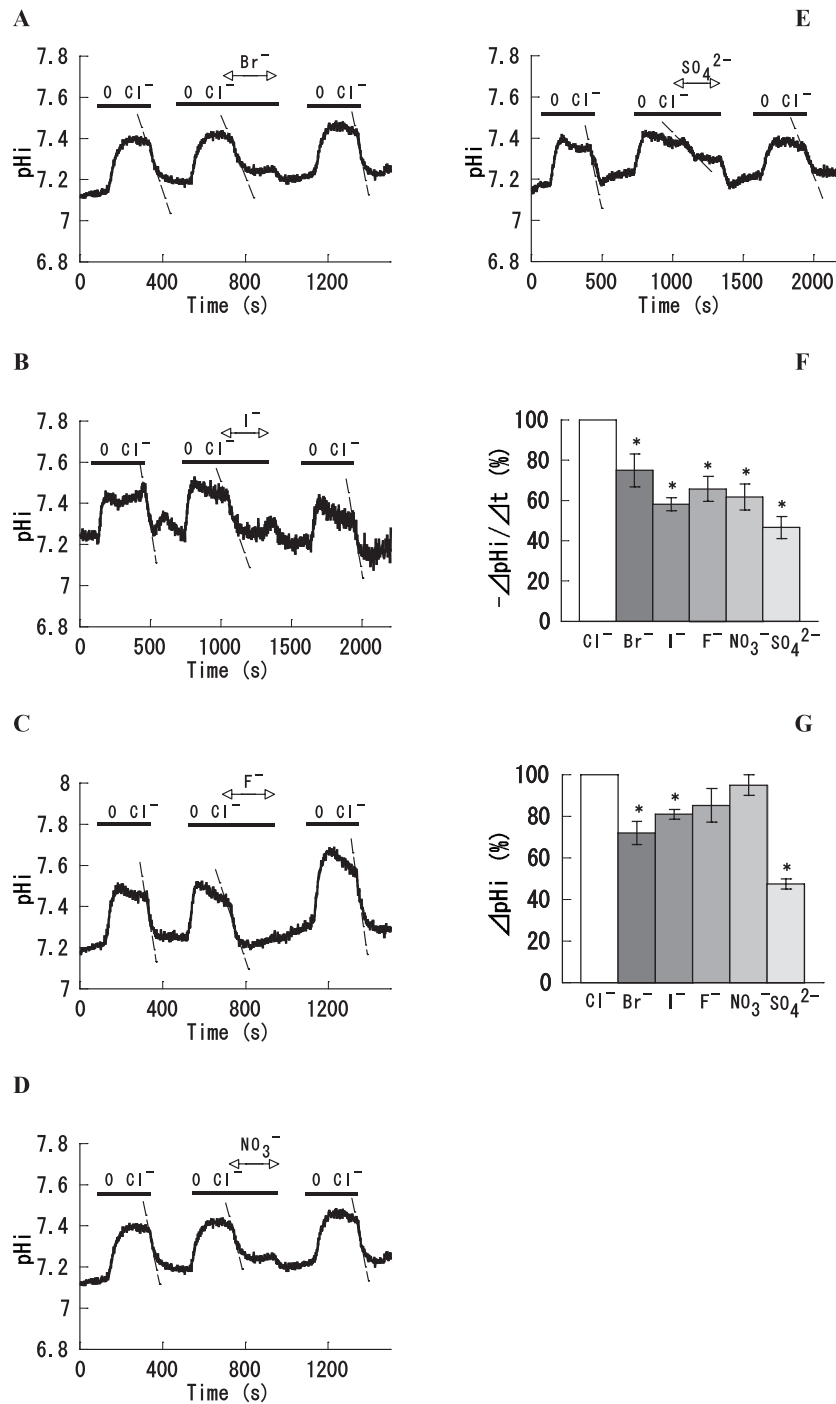
In the next series of experiments, we studied the effect on the  $Cl^-/HCO_3^-$ -exchange activity of several anion transport inhibitors (4). DIDS at a concentration of 30  $\mu M$  slightly but significantly suppressed the rate of  $pH_i$  increase upon  $Cl^-$  removal (about 20%), although the rate of  $pH_i$  recovery upon  $Cl^-$  re-addition was not significantly affected by DIDS

(Fig. 5A). DIDS did not noticeably affect the baseline  $pH_i$  level. Unexpectedly, DIDS at a concentration up to 100  $\mu M$  did not significantly suppress the rate of  $pH_i$  decrease induced by  $Cl^-$  re-addition (Fig. 5B).

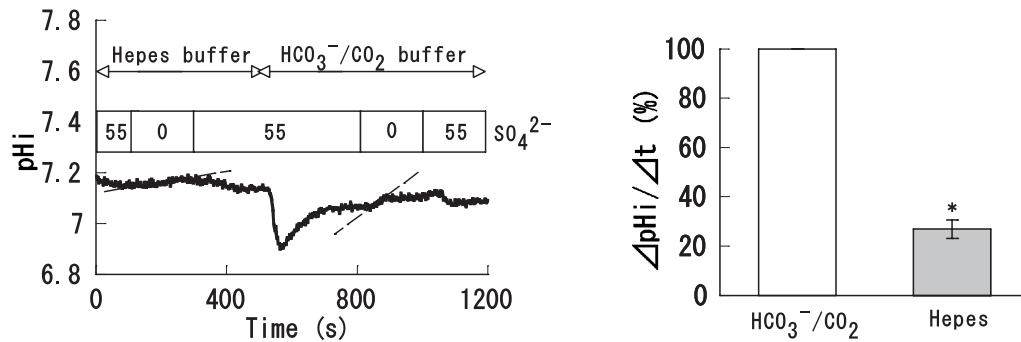
Tenidap (10  $\mu M$ ) and niflumic acid (30  $\mu M$ ), which have been reported to inhibit *slc26a3* (24), significantly inhibited the  $pH_i$  increase upon  $Cl^-$  removal by 30–40% (Fig. 6). NPPB (30  $\mu M$ ), which has been reported to inhibit a variety of anion transporters, including the cystic fibrosis transmembrane conductance regulator (CFTR), significantly suppressed the  $pH_i$  increase upon  $Cl^-$  removal by about 20%.

#### Effect of the high- $K^+$ solution

Some authors have presented evidence for  $Cl^-/HCO_3^-$  exchange mediated by *slc26a6* and *slc26a3* being electrogenic (27, 35), so we examined the effect on the  $Cl^-/HCO_3^-$ -exchange activity of depolarizing the membrane potential by increasing the extracellular  $K^+$  concentration (Fig. 7A). Increasing the  $K^+$  concentration from 5 mM (normal  $K^+$ ) to 60 mM (high  $K^+$ ) caused a reversible increase in the baseline  $pH_i$  value under the  $HCO_3^-/CO_2$ -buffered condition, while it had hardly any effect on the baseline  $pH_i$  value under the  $HCO_3^-/CO_2$ -free, Hepes-buffered condition, indicating that membrane depolarization induced  $HCO_3^-$  uptake and that membrane hyperpolarization stimulated  $HCO_3^-$  release. We then examined the  $Cl^-$  dependence of the membrane hyperpolarization-induced  $HCO_3^-$  release (Fig. 7B). In the absence of  $Cl^-$ , the rate of  $pH_i$  decrease induced by switching from the high- $K^+$  to normal- $K^+$  solution was decreased, but not abolished. Therefore, there were at least two membrane

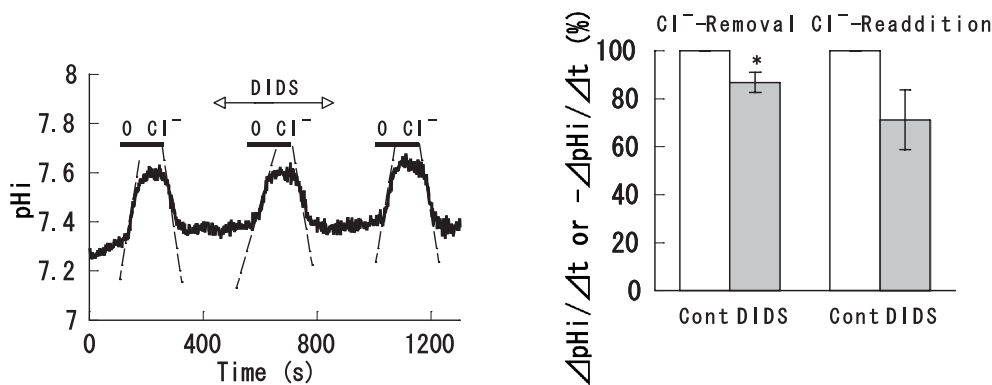


**Fig. 3** Effects of various anions on the HCO<sub>3</sub><sup>-</sup> efflux. The pH<sub>i</sub> decrease induced by Br<sup>-</sup>, I<sup>-</sup>, F<sup>-</sup>, NO<sub>3</sub><sup>-</sup> or SO<sub>4</sub><sup>2-</sup> was determined in the absence of Cl<sup>-</sup>, and compared with that induced by Cl<sup>-</sup>. A-E: Typical recordings. The Cl<sup>-</sup>-removal and re-addition challenge was performed three times on each villus. In the second challenge, the normal Cl<sup>-</sup>-free solution was first switched to the Cl<sup>-</sup>-free solution containing a test anion, and then to the Cl<sup>-</sup>-containing solution. F: the rate of pH<sub>i</sub> recovery induced by each test anion was compared with that induced by Cl<sup>-</sup> (the mean rate of the Cl<sup>-</sup>-induced pH<sub>i</sub> recovery obtained from the first and final challenges). \*p < 0.05, n = 4–6. G: the magnitude of the pH<sub>i</sub> decrease induced by each test anion was compared with that induced by Cl<sup>-</sup> (the mean value of the first and final challenges). \*p < 0.05, n = 4–6. The rate and magnitude of the pH<sub>i</sub> decrease in the final Cl<sup>-</sup>-induced response were not significantly different from those of the first Cl<sup>-</sup>-induced response without regard to the tested anions (data not shown). Br<sup>-</sup>, I<sup>-</sup>, F<sup>-</sup> and NO<sub>3</sub><sup>-</sup> solutions were prepared by substituting 110 mM Na<sup>+</sup> gluconate in the Cl<sup>-</sup>-free solution with 110 mM of each anion (Na salt), while the SO<sub>4</sub><sup>2-</sup> solution was prepared by substituting with 55 mM Na<sub>2</sub>SO<sub>4</sub> and 55 mM mannitol.

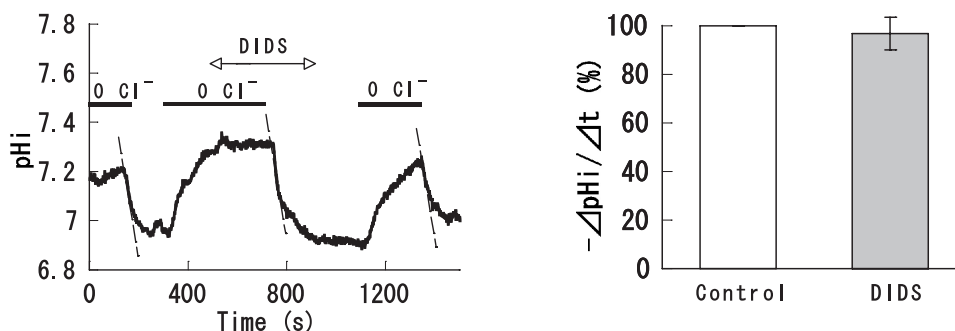


**Fig. 4**  $\text{SO}_4^{2-}$  transport in exchange for  $\text{HCO}_3^-$ . Typical recording (left panel) and a summary (right panel).  $\text{Cl}^-$  was absent throughout the experiment. After the  $\text{pH}_i$  level had been stabilized in the solution containing 55 mM  $\text{SO}_4^{2-}$ , the tissue was challenged first with the  $\text{SO}_4^{2-}$ -free solution in the HEPES-buffered solution and then challenged again in the  $\text{HCO}_3^-/\text{CO}_2$ -buffered solution. The rate of increase in the  $\text{pH}_i$  value upon  $\text{SO}_4^{2-}$  removal was compared between the presence and absence of  $\text{HCO}_3^-/\text{CO}_2$ . \* $p < 0.05$ ,  $n = 4$ .

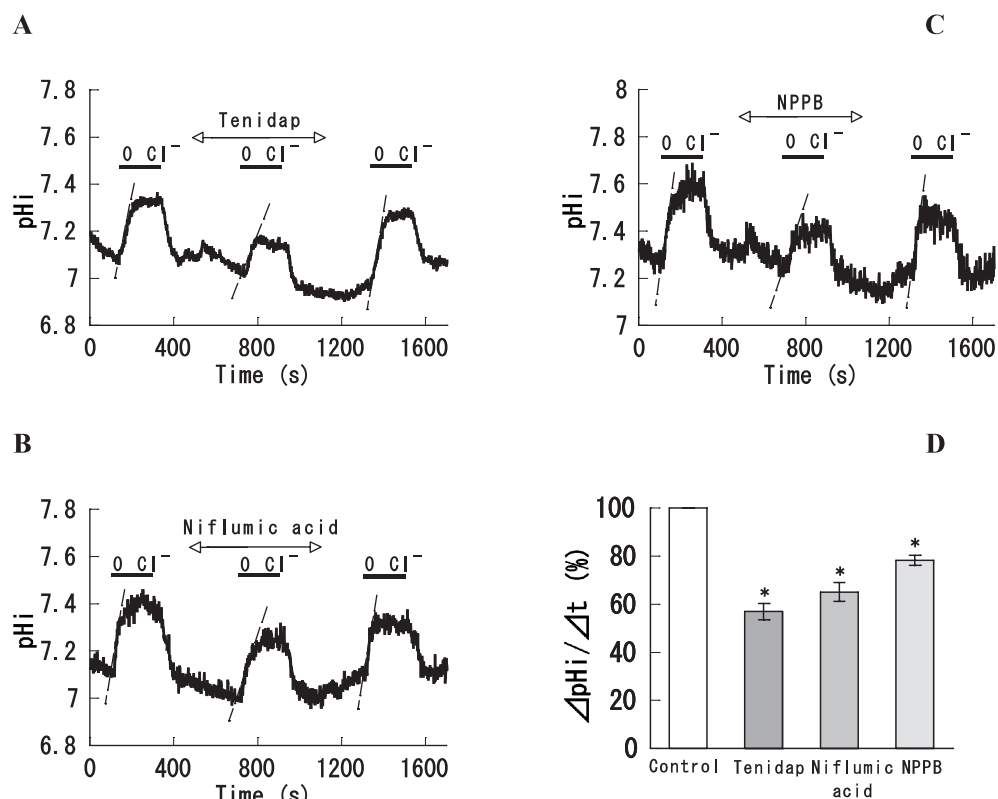
### A 30 $\mu\text{M}$



### B 100 $\mu\text{M}$



**Fig. 5** Effect of DIDS on the  $\text{Cl}^-/\text{HCO}_3^-$ -exchange activity. A: Typical recording (left panel) and summary (right panel) of the effect of 30  $\mu\text{M}$  DIDS. Each tissue was challenged three times by  $\text{Cl}^-$  removal, the first and final challenges being without DIDS and the second challenge being with DIDS. The rates of  $\text{pH}_i$  increase and decrease induced by the  $\text{Cl}^-$ -removal challenge in the presence of 30  $\mu\text{M}$  DIDS (second challenge) was compared with the mean of the first and final control challenges, which were not significantly different from each other (data not shown). \* $p < 0.05$ ,  $n = 4$ . B: Typical recording (left panel) and summary (right panel) of the effect 100  $\mu\text{M}$  DIDS. The rate of  $\text{pH}_i$  decrease in the presence of DIDS was compared with only that observed in the first control  $\text{Cl}^-$ -removal challenge, since the rate observed in the third control challenge appeared to be slightly lower than the rate in the first challenge. This was probably due to slow and irreversible inhibition by a high concentration of DIDS (4).  $n = 4$ .



**Fig. 6** Effects of tenidap, niflumic acid and NPPB on the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity. A-C: Typical recordings. D: Summary of the effects of 10 μM tenidap, 30 μM niflumic acid and 30 μM NPPB. Since the inhibitors caused a decrease in the baseline pHi value (tenidap, 7.29 ± 0.06 to 7.20 ± 0.08, *p* = 0.165; niflumic acid, 7.12 ± 0.01 to 7.04 ± 0.02, *p* < 0.05; NPPB, 7.13 ± 0.06 to 7.05 ± 0.16, *p* < 0.05), the slopes of the pHi increase upon Cl<sup>-</sup> removal were determined at the same pHi level for each tissue. Control values were determined as the mean of the rates obtained in the first and final Cl<sup>-</sup>-removal challenges, these not being significantly different from each other without regard to the tested inhibitors (data not shown). \**p* < 0.05, *n* = 5.

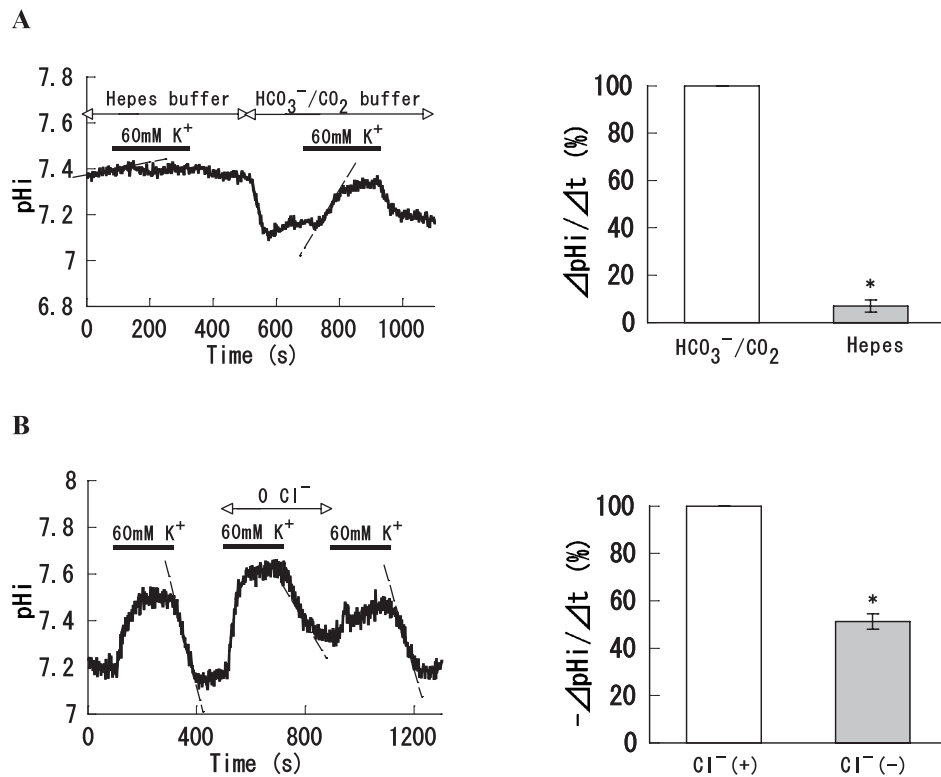
hyperpolarization-induced HCO<sub>3</sub><sup>-</sup> efflux pathways in operation: one was Cl<sup>-</sup>-dependent and the other was Cl<sup>-</sup>-independent.

## DISCUSSION

We initially showed in the present study by RT-PCR that mRNAs for slc26a6, slc26a3 and AE2 were all expressed in the villus of the mouse ileum, confirming the results of previous studies showing that mRNA for these three anion exchangers and proteins for slc26a6 and AE2 were expressed in the mouse small intestine (1, 6, 9, 20, 32). We then demonstrated the presence of Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity in the same preparation. Although previous studies have shown the presence of Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity in villous epithelial cells of the mouse proximal small intestine (duodenum) (29, 33), it remains to be proved whether Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity also occurs in the mouse ileum.

The present results suggest that at least some of the anion exchangers present in the ileal villi can transport not only Cl<sup>-</sup>, but also Br<sup>-</sup>, I<sup>-</sup>, F<sup>-</sup>, SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> in exchange for HCO<sub>3</sub><sup>-</sup> (Figs. 3 and 4). Mouse slc26a6 has been reported to transport Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, Br<sup>-</sup>, I<sup>-</sup> and SO<sub>4</sub><sup>2-</sup> in a heterologously expressed system (7, 12, 35). Mouse AE2 has been shown to transport Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup> and Br<sup>-</sup>, but not I<sup>-</sup>, and SO<sub>4</sub><sup>2-</sup> transport is not known (9). Mouse slc26a3 has been reported to transport NO<sub>3</sub><sup>-</sup> in exchange for HCO<sub>3</sub><sup>-</sup>, but the transport of other anions is not known (26). No matter which transporters are involved, the transport of Cl<sup>-</sup>, Br<sup>-</sup>, I<sup>-</sup>, F<sup>-</sup> and SO<sub>4</sub><sup>2-</sup> across the enterocyte membrane could be responsible for the intestinal absorption of these anions as essential nutrients. This notion may also be applied to the transport of NO<sub>3</sub><sup>-</sup>, because NO<sub>3</sub><sup>-</sup> is contained in food and has been postulated to be absorbed in the small intestine (19).

We have shown that DIDS at a concentration of 30 μM slightly suppressed (by about 20%) the Cl<sup>-</sup>/



**Fig. 7** Effect of the high- $K^+$  solution on the  $Cl^-/HCO_3^-$ -exchange activity. A: Typical recording (left panel) and summary (right panel) of the  $K^+$ -induced  $pH_i$  change. The rates of  $pH_i$  increase induced by switching from the 5 mM  $K^+$ - to 60 mM  $K^+$ -containing solution were compared in the Hepes-buffered and  $HCO_3^-/CO_2$ -buffered solutions. \* $p < 0.05$ ,  $n = 4$ . B: Typical recording (left panel) and summary (right panel) of the  $Cl^-$  dependence of the  $K^+$ -induced  $pH_i$  change in the  $HCO_3^-/CO_2$ -buffered solution. The rates of  $pH_i$  decrease induced by changing the  $K^+$  concentration in the perfusate from 60 mM to 5 mM were compared in the presence and absence of  $Cl^-$ . Since  $Cl^-$  removal shifted the  $pH_i$  value to a higher level, the rates of  $pH_i$  decrease were compared at the same  $pH_i$  level for each tissue. The value in the presence of  $Cl^-$  was the mean of the first and third measurements which were not significantly different from each other (data not shown). \* $p < 0.05$ ,  $n = 10$ .

$HCO_3^-$  exchange activity, but that increasing the DIDS concentration up to 100  $\mu M$  did not further suppressed it (Fig. 5). Mouse *slc26a3* has been shown to be inhibited by 1 mM DIDS by only 24% in a heterologously expressed system, indicating that its relative resistance to DIDS (20). In contrast,  $Cl^-/HCO_3^-$  exchange mediated by mouse *slc26a6* was inhibited by 5  $\mu M$  DIDS by about 90% (7, 12, 27). AE2 has also been reported to be sensitive to DIDS: the  $Cl^-$  uptake mediated by mouse AE2 and the  $Cl^-/HCO_3^-$  exchange mediated by rat AE2 were inhibited by DIDS with an  $ED_{50}$  of less than about 20  $\mu M$  (9, 18). It is not clear why 100  $\mu M$  DIDS did not affect the  $Cl^-/HCO_3^-$  exchange activity (Fig. 5B), although a possible explanation is that a high concentration of DIDS had a stimulatory effect on the exchange activity in addition to the inhibitory effect, and that both effects cancelled each other.

We found that niflumic acid, tenidap and NPPB each partially inhibited  $Cl^-/HCO_3^-$  exchange. Niflu-

mic acid and tenidap have been reported to inhibit mouse *slc26a6* (7) and human *SLC26A3* (6). However, the sensitivity of mouse *slc26a6* to tenidap and NPPB, and of mouse *slc26a3* and AE2 to niflumic acid, tenidap and NPPB is not known.

We found that increasing or decreasing the extracellular  $K^+$  concentration respectively induced an uptake or release of  $HCO_3^-$ . Since the membrane  $K^+$  channels are ubiquitously expressed, respectively increasing or decreasing the extracellular  $K^+$  concentration may well depolarize or hyperpolarize the membrane potential, thereby driving  $HCO_3^-$  transport if it occurs through an electrogenic pathway. A portion of this  $K^+$ -induced  $HCO_3^-$  transport was  $Cl^-$  dependent (Fig. 7). The  $Cl^-$ -dependent, electrogenic  $HCO_3^-$  transport pathway may be mediated by *slc26a6*, because mouse and human *slc26a6* has been suggested to mediate electrogenic  $Cl^-/HCO_3^-$  exchange with stoichiometry of  $HCO_3^- > Cl^-$  (14, 27, 35), although some authors have provided evi-



dence against the electrogenicity of SLC26A6 (7). Slc26a3 may also be electrogenic, but could not mediate this K<sup>+</sup>-induced HCO<sub>3</sub><sup>-</sup> transport since it has been suggested to operate with stoichiometry of HCO<sub>3</sub><sup>-</sup> < Cl<sup>-</sup> (6, 14, 16, 27). AE2 is probably electroneutral (26), excluding the possibility that this is involved in mediating K<sup>+</sup>-induced HCO<sub>3</sub><sup>-</sup> transport.

One possible mechanism to explain Cl<sup>-</sup>-independent, K<sup>+</sup>-induced HCO<sub>3</sub><sup>-</sup> transport is an HCO<sub>3</sub><sup>-</sup>-permeable channel. CFTR is one candidate for such a channel and has been reported to be present in mouse villous enterocytes of the proximal small intestine (3), while also being known to permeate HCO<sub>3</sub><sup>-</sup> (24, 25). It is also possible that Cl<sup>-</sup>-independent, K<sup>+</sup>-induced HCO<sub>3</sub><sup>-</sup> transport is mediated by the K<sup>+</sup>-HCO<sub>3</sub><sup>-</sup> co-transporter. Such a co-transporter has previously been suggested to be present in squid axon (37) and in the medullary thick ascending limb of rat kidney (34), although its molecular identity remains to be determined. Since K<sup>+</sup>-induced HCO<sub>3</sub><sup>-</sup> transport has also been observed in the villous cells of the mouse proximal small intestine (29) and guinea pig distal small intestine (Hayashi, unpublished observation), it may be a widely-distributed phenomenon and, therefore, an important issue to be addressed further.

In summary, it is at least possible that slc26a6, slc26a3 and AE2 are all involved in the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange activity in mouse ileal villous cells. slc26a6 in particular may be involved, since SO<sub>4</sub><sup>2-</sup>/HCO<sub>3</sub><sup>-</sup> exchange activity was demonstrated. The inhibition of Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange by niflumic acid and DIDS also agrees with the functional expression of slc26a6. The possible presence of electrogenic Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange with stoichiometry of Cl<sup>-</sup> < HCO<sub>3</sub><sup>-</sup> could also support the involvement of slc26a6, although the relatively small inhibition of Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange by DIDS may exclude any substantial involvement of slc26a6. slc26a3 may also be implicated in the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange activity, since a substantial portion of this exchange was DIDS-insensitive, while being inhibited by tenidap and niflumic acid. AE2 may also mediate Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange activity judging from the inhibition by DIDS. We have considered mainly the studies on mouse exchangers in this discussion, and not on human orthologues, since important differences have been demonstrated for SLC26A6 between the mouse and human (7). Several other Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchangers, *i.e.*, AE3 and AE4, have been suggested to be present in intestinal epithelial cells (2, 15, 36), and thus could also be responsible for the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup>-exchange activity in the mouse ileum that was ob-

served here. In addition, it is by no means inevitable that the HCO<sub>3</sub><sup>-</sup> transport induced by Cl<sup>-</sup> removal would be due to Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchangers, because it could also be the result of indirect, electrical coupling when Cl<sup>-</sup> and HCO<sub>3</sub><sup>-</sup> are both transported through the anion channels (3, 24, 25).

Previous studies on villous epithelial cells from the mouse proximal small intestine (duodenum) have demonstrated the major role of slc26a6 in the Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange activity (29, 33). The proximal and distal parts of the small intestine may have distinctly different physiological functions from each other and, therefore, it would be interesting to further explore the difference in enterocyte Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> exchange between the proximal and distal small intestine.

#### Acknowledgement

We thank Tony Innes of Link Associates for helping us to edit the English text.

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