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## Simulations of Bulk Effects of Zr on Cohesion of Ni<sub>3</sub>Al Grain Boundary\*

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**摘要:** Monte Carlo simulation with the energetics described by the embedded atom method has been employed to calculate dependence of the grain boundary cohesion on the Zr bulk atom concentration, for the Ni<sub>3</sub>Al- $x$  % Zr (100% Ni\100% Ni)[001]/ $\Sigma$ 5(210)/36.87° symmetric tilt grain boundary, at the equilibrium. Calculations show that when  $x$  (the Zr bulk atom concentration) increases from 0.1 to 0.5, the Zr enrichment increases, both the Ni enrichment and the Al depletion maximizes at  $x = 0.3$ . The calculations also show the best cohesion of the grain boundary at  $x = 0.3$ .

**Key words:** grain boundary concentration; inducing and segregating specie; bulk effect

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### 1 Introduction

The intermetallic compound Ni<sub>3</sub>Al has high strength at elevated temperature with a characteristic positive temperature dependence on strength over a certain temperature range<sup>[1]</sup>. This remarkable property, however, is not sufficient to make it useful for high-temperature applications because of its brittleness and unfabricability. It has been demonstrated in experiments that the addition of a small amount of B, Cr, Mg or Zr (trace element) to Ni<sub>3</sub>Al remarkably improves its ductility<sup>[1-9]</sup>.

Quantitative dependence of the cohesive properties of the Ni<sub>3</sub>Al grain boundaries (or ductility of Ni<sub>3</sub>Al) on the bulk concentration of the trace element has been studied for years. For example, for

the Ni<sub>3</sub>Al-B system, the measurements<sup>[3]</sup> and the simulations<sup>[4, 5, 9]</sup> have shown that the cohesive properties of the Ni<sub>3</sub>Al grain boundaries will be improved when its B bulk atom concentration is below about 1 %, and also that the best cohesive properties of the Ni<sub>3</sub>Al grain boundaries will be obtained when its B bulk atom concentration equals about 0.5 % (weight fraction 0.1 %). However, for the Ni<sub>3</sub>Al-Zr system, to our knowledge, this dependence has been not studied until now. The aim of present simulation work studies such dependence. The simulation method is the Monte Carlo one conjoined with the embedded atom method (EAM) po-

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tentials<sup>[4-9]</sup>.

## 2 EAM Potentials and Monte Carlo Simulation

### 2.1 Potentials

The EAM energy of an  $n$ -particle homonuclear system is given by

$$E = \frac{1}{2} \sum_{i \neq j}^n \phi(r_{ij}) + \sum_i^n F(\bar{\rho}_i), \quad (1)$$

where  $E$  is the total energy of the system,  $r_{ij}$  is the distance between atom  $i$  and  $j$ , and  $\phi(r)$  is a pairwise interaction potential which is taken to be a Morse function,  $F(\rho)$  is the embedding energy and  $\bar{\rho}_i$  is the electron density at atom  $i$ , which is contributed by its all neighbors. For the Ni<sub>3</sub>Al-Zr system, the optimized potential parameters are adopted from Ref. [8].

### 2.2 Monte Carlo simulation

Monte Carlo method is used to relax the grain structure to the lowest energy configuration at the room temperature (25°C). The unrelaxed, starting structure of the grain boundary is assumed to be the coincidence site lattice (CSL) model. The (100%Ni\100%Ni)[001]/Σ5(210)/36.87° symmetric tilt boundary structure of Ni<sub>3</sub>Al is chosen.

A schematic drawing of the computation cell is shown in Fig. 1 of Ref. [5]. The initial bulk concentration of Zr atoms in the computation cell is given as  $x$  atoms % ( $x = 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9$  and  $1.0$ ), and this concentration disperses randomly homogeneously. The present results in Sect. 3 are statistical (average of a few million MC steps per atom after equilibrium) for each system of 82 620 atoms.

## 3 Results and Discussion

### 3.1 Definition of the bulk effects of Zr

Starting with relaxed Ni<sub>3</sub>Al-Zr grain boundary, as the segregating species, the Zr atoms substitute either Al or Ni atoms<sup>[8]</sup>; in our opinion,

meantime, as the inducing species, the Zr atoms induce Ni atoms to substitute Al atoms. Subsequently, the grain boundary is relaxed to the lowest energy. Thus, the bulk effects of Zr on the grain boundary include the primarily positive effect on the Ni enrichment at the grain boundary, the secondarily positive effect on it, and the negative effect on it. The primarily positive effect is defined as the Zr atoms inducing Ni atoms to substitute into Al sites, the secondarily positive effect is defined as the Zr atoms substituting into Al sites, and the negative effect is defined as the Zr atoms substituting into Ni sites.

### 3.2 Analysis of the bulk effects on boundary concentrations

For the Ni<sub>3</sub>Al- $x$  atoms % Zr (100%Ni\100%Ni)[001]/Σ5(210)/36.87° symmetrical tilt grain boundary, at the equilibrium, Fig. 1 shows that when  $x$  (the Zr bulk concentration) increases from 0.1 to 0.5, the Zr enrichment increases, both the Ni enrichment and the Al depletion maximizes at  $x = 0.3$ .

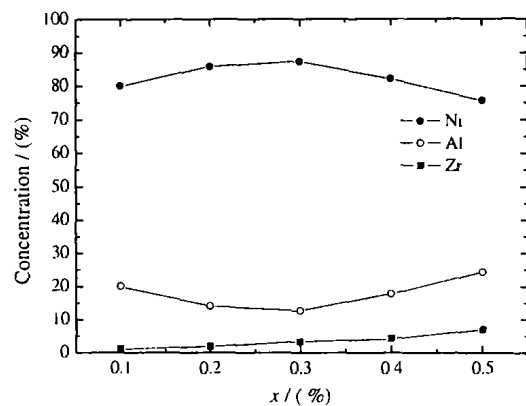


Fig. 1 The atom concentrations of Zr, Ni and Al as a function of  $x$  (the Zr bulk atom concentration) at the equilibrium for the Ni<sub>3</sub>Al- $x$  % Zr [001]/Σ5(210)/36.87° symmetrical grain boundary.

For relaxation of the Ni<sub>3</sub>Al- $x$  atoms % Zr grain boundary, in our opinion, the primarily positive effect is main cause of the Ni enrichment at the grain boundary, the secondarily positive effect plays an “enhancing” role; but the negative effect

plays a “weakening” role, especially at high Zr bulk concentrations such as  $x = 0.5$ , where this negative effect becomes strong, because the significant Zr enrichment (Fig. 1) is formed by the 100% Zr atoms substituting into Ni sites (Fig. 2).

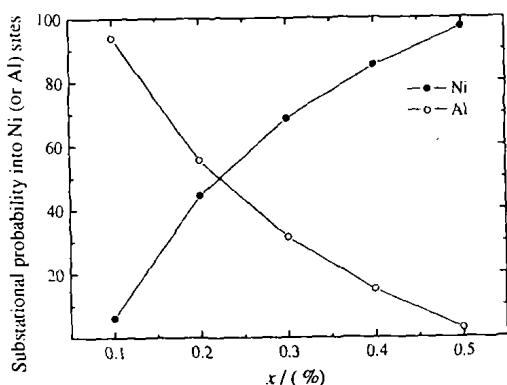


Fig. 2 The substitutional probability of Zr atoms into Al (or Ni) sites as a function of  $x$  (the Zr bulk atom concentration) at the equilibrium for the  $\text{Ni}_3\text{Al-}x\%$  Zr  $[001]/\Sigma 5(210)/36.87^\circ$  symmetrical grain boundary.

Also in our opinion, study of combination of the positive and the negative effects is available, in order to understand the dependence of the cohesive properties of  $\text{Ni}_3\text{Al}$  grain boundaries on the Zr bulk concentration, particularly in order to search the Zr bulk concentration corresponding to the best cohesive properties of  $\text{Ni}_3\text{Al}$  grain boundaries.

According to the above explanations, for the  $\text{Ni}_3\text{Al-}x\%$  Zr grain boundary at the equilibrium, at low Zr bulk concentrations, such as  $x = 0.1$ , due to the positive effects, corresponding to the Zr enrichment increasing, the Ni enrichment and the Al depletion increase (Fig. 1). However, at high Zr bulk concentrations, such as  $x = 0.5$ , due to the negative effect, corresponding to the Zr enrichment increasing, the Ni enrichment and the Al depletion decrease (Fig. 1). Therefore, between low ( $x = 0.1$ ) and high ( $x = 0.5$ ) Zr bulk concentrations, there might be a Zr bulk concentration, at which the Ni enrichment and the Al depletion maximize ( $x = 0.3$  (Fig. 1)).

### 3.3 Analysis of the bulk effects on cohesive properties

As shown in Fig. 1, for the  $\text{Ni}_3\text{Al-}x\%$  Zr grain boundary, at the equilibrium, when  $x$  (the Zr bulk concentration) increases from 0.1 to 0.5, the Zr enrichment increases, both the Ni enrichment and the Al depletion maximizes at  $x = 0.3$ . Are the best cohesive properties of the  $\text{Ni}_3\text{Al-}x\%$  Zr grain boundary obtained at  $x = 0.3$ ? Calculations of the Griffith grain boundary cohesion energy ( $\gamma_{\text{coh, for GB (210)}}$ ) and the bulk one ( $\gamma_{\text{coh, for bulk (210)}}$ ) are necessary to answer this question<sup>[9]</sup>.

For the  $\text{Ni}_3\text{Al-}x\%$  Zr (210) bulk, the present calculations show that  $\gamma_{\text{coh, for bulk (210)}}$  ( $\cong 4\,327\text{ mJ/m}^2$  in the range of  $x = 0.1-0.5$ , i. e., its relative change is always less than 0.5% ( $-0.5\% < (1 - \gamma_{\text{coh, for bulk (210)}) / 4\,327\text{ mJ/m}^2) \times 100\% < 0.5\%$ ) when  $x$  changes between 0.1 and 0.5. For the  $\text{Ni}_3\text{Al-}x\%$  Zr (210) grain boundary, the present calculations show that  $\gamma_{\text{coh, for GB (210)}} / \gamma_{\text{coh, for bulk (210)}} > 0.79$  in the range of  $x = 0.1-0.5$ , and its best value is 0.89 at  $x = 0.3$ , i. e., the best cohesion of the grain boundary is obtained when  $x = 0.3$  (Fig. 3). Note that for the  $\text{Ni}_3\text{Al}$  grain boundary (210),  $\gamma_{\text{coh, for GB (210)}} / \gamma_{\text{coh, for bulk (210)}} = 0.73$ .

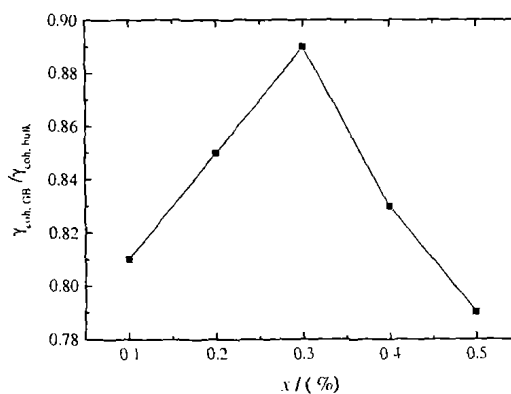


Fig. 3 The Griffith cohesive energy ratio of the GB (grain boundary) to the bulk, i. e.  $\gamma_{\text{coh, for GB}} / \gamma_{\text{coh, for bulk}}$ , as a function of  $x$  (the Zr bulk concentration) for the  $\text{Ni}_3\text{Al-}x\%$  Zr  $[001]/\Sigma 5(210)/36.87^\circ$  symmetrical grain boundary.

## 4 Conclusion

Monte Carlo simulation with the energetics described by the embedded atom method (EAM)

has been employed to calculate not only grain boundary concentrations of Ni, Zr and Al, but also dependence of the grain boundary cohesion on the Zr bulk concentration, for the  $\text{Ni}_3\text{Al}-x\% \text{Zr}$  ( $100\% \text{Ni} \setminus 100\% \text{Ni}$ ) $[001]/\Sigma 5(210)/36.87^\circ$  symmetric tilt grain boundary, at the equilibrium. The

calculations show that when  $x$  (the Zr bulk atom concentration) increases from 0.1 to 0.5, the Zr enrichment increases, both the Ni enrichment and the Al depletion maximizes at  $x=0.3$ . The calculations also show the best cohesion of the grain boundary at  $x=0.3$ .

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## 模拟锆(Zr)的基体效应对 $\text{Ni}_3\text{Al}-x \text{ atoms } \%$ Zr 晶界内聚性的影响\*

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**摘 要:** 利用双粒子模型研究微量元素锆(Zr)的基体效应对  $\text{Ni}_3\text{Al}-x \text{ atoms } \%$  Zr 晶界内聚性的影响。模型显示,  $x$ [锆(Zr)的基体浓度]从 0.1 增加到 0.5, 在晶界 Zr 富集是增加的; Ni 富集和 Al 贫乏在  $x=0.3$  时趋于最大。模型还显示,  $\text{Ni}_3\text{Al}-x \text{ atoms } \%$  Zr 晶界的内聚性在  $x=0.3$  时为最佳。

**关键词:** 晶界浓度; 偏析子与诱发子; 基体效应