

Chemical Constituents of *Linaria aucheri*

Dilek ERCİL*, M. Koray SAKAR

Hacettepe University, Faculty of Pharmacy, Department of Pharmacognosy,
06100 Ankara, TURKEY
e-mail: dercil@hacettepe.edu.tr

Esther DEL OLMO, Arturo SAN FELICIANO

University of Salamanca, Faculty of Pharmacy, Department of Organic and
Pharmaceutical Chemistry, 37007 Salamanca, SPAIN

Received 10.06.2003

From the petroleum ether and methanol extracts of the aerial parts of *Linaria aucheri* (Scrophulariaceae) 6 known compounds: β -amyirin (**1**), ergost-7-en-3 β -ol (**2**), stigmasta-5,22-*E*-dien-3 β -ol (**3**), stigmast-5-en, 24*S*-3 β -ol (**4**), antirrinoside (**5**) and linariin (**6**), were isolated and identified.

The structures of these compounds have been elucidated by spectroscopic methods (UV, IR, ^1H NMR, ^{13}C NMR, DEPT-135 and GC-MS).

Key Words: *Linaria aucheri*, Scrophulariaceae, lypophylic compounds, β -amyirin, antirrinoside, linariin.

Introduction

The genus *Linaria* (Scrophulariaceae) is represented by 20 species in the flora of Turkey¹. Several species have been used in traditional medicine as tonics, antiscorbutics, laxatives, antidiabetics and diuretics, as well as for the treatment of wounds, hemorrhoids and vascular disorders^{2,3}.

Previous studies of *Linaria* species have shown the presence of flavonoids and their glycosides, as well as ionol glucosides, iridoids, alkaloids, diterpenoids and phenylethanoids³⁻¹². In the present paper, we report the initial isolation and identification of some known lypophylic compounds, together with an iridoid and flavon glycosides, from the aerial parts of *Linaria aucheri* petroleum ether and methanol extracts.

Experimental

General Experimental Procedures: UV spectra were obtained in MeOH on a Shimadzu spectrophotometer UV 160A. IR spectra were recorded on a Perkin Elmer spectrophotometer FTIR 1720X. The ^1H and ^{13}C spectra were obtained on a Bruker WP 200 SY spectrometer at 200 and 50.29 MHz, respectively. Silica gel 60 (0.063-0.200 mm, 7734) for column chromatography was purchased from Merck. Vacuum-liquid

*Corresponding author

chromatography (VLC) was performed on a glass column (2.2 x 15 cm) filled with RP-18 (20-45 μm , LiChroprep) material. GC-MS apparatus: Hewlett-Packard 5890 Series II. Ion Sourced: 70 eV electron-impact ion source. Column used: SPB-1, methylsilicone, 12 m x 0.2 mm internal diameter, 0.33 μm width wall. Initial temperature: 100 °C, 5 °C increase per min. Final temperature: 300 °C for 5 min. Carrier gas: He.

Plant Material: *Linaria aucheri* Boiss. (Scrophulariaceae) was collected from Baykuşboğazı, Çankırı (Northeast Anatolia) in July 1997. A voucher specimen has been deposited in the Herbarium of the Faculty of Pharmacy, Hacettepe University (HUEF 97-020).

Extraction and Isolation: The air-dried and powdered aerial parts of *L. aucheri* (950 g) were extracted successively with petroleum ether in a mechanical stirrer (40-60 °C) (3 x 2.5 l) and methanol (2 x 1.5 l). The petroleum ether extracts were evaporated to dryness in vacuo (10 g) and then defatted with acetone (6 x 100 mL). The crude extract (4.5 g) was subjected to the Si gel column (3 x 40 cm) with a gradient elution of 100% hexane to 25% ethyl acetate (1 l; 20 mL, each), to give 2 distinctive fractions (AI and AII). Fraction AI (Fr. 27-32, 338 mg) was purified by preparative TLC (5554 Merck) with a mixture of cyclohexane-ethyl acetate (8:2) to yield **1** (17 mg). Fraction AII (Fr. 34-48, 80 mg) is a mixture of compounds **2-4**, with a ratio of 1:1:2, respectively.

The methanolic extract (8 g) was subjected to RP-18 [20-45 μm , LiChroprep C-18 (Merck)] VLC with a gradient elution of methanol-water (10% to 90% MeOH) mixture. This yielded 5 main fractions (Frs. BI-BV, each of 100 mL). Fraction BII (1300 mg) yielded an iridoid (**5**), whereas fraction BV (110 mg) yielded a flavonoid (**6**).

Acid Hydrolysis of 5 and 6: Compound **5** was applied to a TLC plate (Silica gel 60 F₂₅₄, 0.2 mm, Merck) and the plate was treated with concentrated HCl vapor in a closed tank for 1 h. After the evaporation of the concentrated HCl, authentic sugar samples were applied to the TLC plate, and the plate was developed in a EtOAc-MeOH-conc.HOAc-H₂O (60:15:15:10, v/v) solvent system. Spots were visualized by spraying with a Thymol-EtOH-conc.H₂SO₄ (0.5 g:95 mL:5 mL) reagent and heated at 110 °C for 5 min. The same procedure was used for compound **6**, though this was treated for 2 h with concentrated HCl vapor. The sugars were identified as glucose for compound **5**, and glucose and rhamnose for compound **6**.

Results and Discussion

β -Amyrin (1): UV λ_{max} (MeOH): 240.2 nm; IR γ_{max} (KBr) cm^{-1} : 3397, 2932, 1645, 1465, 1379, 1029, 900; ¹H NMR (200 MHz, CDCl₃): δ 5.12 (m, H-12), 3.23 (m, H-3), 1.13, 0.99, 0.97, 0.94, 0.87 (x 2), 0.83, 0.79 (CH₃). ¹³C NMR (50.29 MHz, CDCl₃): δ 38.14 (C-1), 27.51 (C-2), 79.12 (C-3), 38.88 (C-4), 55.27 (C-5), 18.44 (C-6), 33.03 (C-7), 38.82 (C-8), 47.81 (C-9), 37.00 (C-10), 23.48 (C-11), 121.82 (C-12), 145.28 (C-13), 42.18 (C-14), 26.12 (C-15), 27.37 (C-16), 32.04 (C-17), 47.22 (C-18), 46.93 (C-19), 31.34 (C-20), 34.83 (C-21), 37.26 (C-22), 28.22 (C-23), 15.47 (C-24), 15.72 (C-25), 16.97 (C-26), 25.56 (C-27), 28.44 (C-28), 33.44 (C-29), 23.48 (C-30); EI-MS: m/z 426 (M⁺, 0.6%), 411 (0.1%), 218 (100%), 272 (0.1%), 189 (30%), 135 (34%), 95 (48%).

Ergost-7-en-3 β -ol (2): EI-MS: m/z 400 (M⁺, 100%), 382 (43%), 367 (30%), 273 (32%), 255 (36%), 161 (48%), 147 (40%), 107 (77%), 43 (84%).

Stigmasta-5,22E-dien-3 β -ol (3): EI-MS: m/z 412 (M⁺, 66%), 369 (11%), 300 (36%), 271 (55%),

255 (56%), 159 (60%), 147 (43%), 133 (58%), 97 (54%), 83 (100%), 55 (98%).

Stigmast-5-en, 24S-3 β -ol (4): EI-MS: m/z 414 (M⁺, 98%), 381 (26%), 329 (52%), 303 (50%), 273 (33%), 213 (54%), 145 (76%), 119 (61%), 107 (88%), 95 (81%), 43 (100%).

Antirrinoside (5): UV λ_{max} (MeOH) nm: 212, 232 sh; IR γ_{max} (KBr) cm⁻¹: 3392 (OH), 2922 (=C-H), 1657 (C=C), 1402, 1231, 1014, 960, 894, 860; ¹H and ¹³C NMR (see Table 1).

Table 1. ¹H and ¹³C NMR spectroscopic data of **5** (δ ppm, CD₃OD).

C/H atom	δ_H	δ_C
Aglycone		
1	5.31 (d, $J = 7.3$ Hz)	95.0
3	6.30 (d, $J = 6.4$ Hz)	142.9
4	4.80 (d, $J = 6.4$ Hz)	107.6
5	-	74.5
6	3.88 (d, $J = 1.6$ Hz)	78.1
7	3.31 (d, $J = 1.6$ Hz)	66.2
8	-	64.1
9	2.30 (br.d, $J = 7.3$ Hz)	53.0
10	1.38 (3H,s)	17.6
Glucose		
1'	4.59 (d, $J = 7.3$ Hz)	99.5
2'	3.09-3.26*	74.5
3'	3.09-3.26*	78.3
4'	3.09-3.26*	71.6
5'	3.09-3.26*	77.5
6'a	3.83 (dd, $J = 1.8/11.7$ Hz)	62.8
6'b	3.52 (dd, $J = 6.2/11.7$ Hz)	

*Signal pattern unclear due to overlapping

Linariin (6): UV λ_{max} (MeOH) nm: , 230sh, 277, 328; (NaOMe) nm: 294.5; (AlCl₃) nm: 230.5sh, 288.5, 300, 355; (AlCl₃+HCl) nm: 230.5sh, 289.5, 299, 353; (NaOAc) nm: 276, 328; (NaOAc+H₃BO₃) nm: 276.5, 328.5; IR γ_{max} (KBr) cm⁻¹: 3402, 2916, 1723, 1688, 1610, 1583, 1514, 1490, 1429, 1301, 1189, 835. ¹H and ¹³C NMR (see Table 2).

The ¹H NMR spectrum of compound **1** indicated the presence of 1 vinylic proton (δ 5.12, t) and 8 methyl groups (δ 0.99, H-23; 0.79, H-24; 0.94, H-25; 0.97, H-26; 1.13, H-27; 0.83, H-28; 0.87, H-29, H-30). The structure of **1** is supported by the appearance of 8 methyl, 10 methylene, 5 methine and 7 quaternary carbons of 30 signals in the ¹³C NMR and DEPT-135 spectra.

The EIMS spectrum of compound **1** showed the molecular ion at m/z 426, consistent with the formula C₃₀H₅₀O. The structure of **1** was identified by GC-MS. The IR and NMR (¹H, ¹³C and DEPT-135) spectra, as well as the GC-MS data, permitted the structure of **1** to be assigned as β -amyrin¹³⁻¹⁵.

The remaining compounds were identified as ergost-7-en-3 β -ol (**2**), stigmasta-5,22-*E*-dien-3 β -ol (**3**) and stigmast-5-en, 24S-3 β -ol (**4**) by direct comparison of their GC-MS data with those given in the literature¹⁶⁻²¹.

Table 2. ^1H and ^{13}C NMR spectroscopic data of **6** (δ ppm, CD_3OD).

H/C atom	δ_H	δ_C
Aglycone		
2	-	103.8
3	6.80 (1H, s)	182.8
4	-	
5-OH	12.90 (1H, br.s)	152.8
6	-	133.1
7	-	152.8
8	6.91 (1H, s)	94.9
9	-	156.9
10	-	106.4
1'	-	123.3
2'/6'	7.97 (d, $J = 9.0$ Hz)	128.8
3'/5'	7.08 (d, $J = 9.0$ Hz)	115.1
4'	-	163.0
6-OMe	3.88 (3H, s)	60.6
4'-OMe	3.82 (3H, s)	55.7
Glucose		
1''	5.12 (d, $J = 7.3$ Hz)	100.6
2''	3.2-3.9 (6 H, m)*	73.7
3''	3.2-3.9 (6 H, m)*	76.9
4''	3.2-3.9 (6 H, m)*	69.7
5''	3.2-3.9 (6 H, m)*	75.9
6''	3.2-3.9 (6 H, m)*	65.9
Rhamnose		
1'''	4.59 (1H, br.s)	100.5
2'''	3.2-3.9 (3 H, m)*	70.8
3'''	3.2-3.9 (3 H, m)*	68.8
4'''	4.69 (t, $J = 9.8$ Hz)	74.3
5'''	3.2-3.9 (3 H, m)*	66.3
6'''	0.82 (d, $J = 6.2$ Hz)	17.4
-OCOCH ₃	1.93 (3H, s)	20.9
-OCOCH ₃		170.6

*Signal pattern unclear due to overlapping

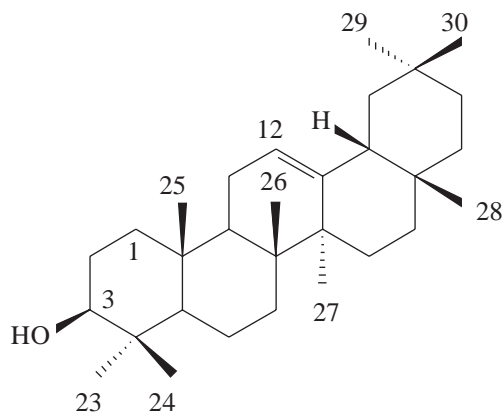


Figure 1. β -Amyrin (**1**).

Substance **5**, which is 1 of the 2 major compounds, is an iridoid glucoside. The UV spectrum of **5**

showed the characteristic absorption bands for a non-conjugated iridoid enol-ether system at 212 and 232(sh) nm (MeOH). In the IR spectrum, it exhibited a distinctive absorption band of an enol-ether function at 1657 cm^{-1} , as well as a broad hydroxyl absorption band at 3392 cm^{-1} . The ^1H , ^{13}C NMR and DEPT-135 (Table 1) spectra showed that compound **5** was an hydroxylated epoxy iridoid glucoside. The anomeric proton signal at δ 4.59 (d, $J = 7.3$ Hz), as well as the doublets at δ 3.83 ($J = 1.8/11.7$ Hz) and 3.52 ($J = 6.2/11.7$ Hz), were assigned to a β -oriented glucopyranose moiety in **5**. On the other hand, the ^{13}C NMR and DEPT-135 data of the related carbon atoms also supported the presence of a β -glucopyranose in compound **5**. Hydroxyl groups of the aglycon appeared to be located at C-5 and C-6, owing to the broad doublet at δ 4.80, which was assigned to the H-4 proton that coupled with H-3 ($J_{3,4} = 6.4$ Hz) and H-9 protons ($J_{4,9} < 1$ Hz). This assumption was also proved by the chemical shift values of C-5 (δ 74.5) and C-6 (δ 78.1). The ^1H and ^{13}C NMR resonances at C-7 and C-8 positions were indicative of the presence of an epoxy-system between C-7 and C-8, as a result of comparing these values with those analogues containing a similar epoxy-system at C-7 and C-8^{22,23}. A singlet at δ 1.33 (3H) was assigned to the methyl group in the α position, which was assumed to be placed at C-8 (δ 64.1, s). Concerning the stereochemistry of the hydroxyl group at C-6, the chemical shift value of $J_{6,7}$ ($J = 1.6$ Hz) is indicative of a *cis*- configuration between H-6 and H-7. The H-9 proton was observed as a broad doublet ($J = 7.3$ Hz) due to the coupling between H-1 and H-9. Based on its spectroscopic data as well as a comparison with those published in the literature, the structure of **5** was established as anthirrinoside^{7-9,23,24}.

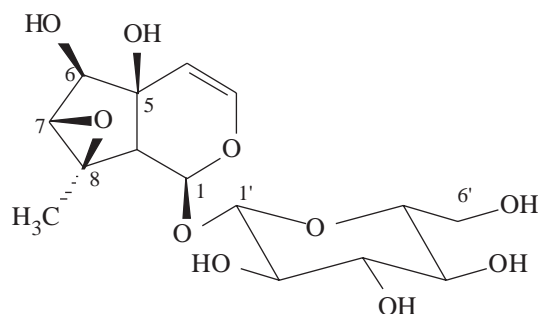


Figure 2. Anthirrinoside (**5**).

The absorption maxima at 328 and 277 nm in the UV spectrum of compound **6** (MeOH) were attributed to a flavone skeleton. The absence of a bathochromic shift in the UV spectrum with the presence of sodium acetate showed the glycosylation of the 7-OH group of the aglycon. In the IR spectrum of **6**, the absorption band at 1730 cm^{-1} was characteristic of an ester group.

The signals in the ^1H , ^{13}C NMR and DEPT-135 spectroscopic data of **6** (Table 2) supported the presence of a flavone skeleton. The ^1H NMR signals at δ 5.12 (d, $J = 7.3$ Hz) and 4.59 (br s) were attributed to the anomeric protons of a β -glucopyranose and α -rhamnopyranose units, respectively. In addition, the methyl signal at δ 0.82 (d, $J = 6.2$ Hz) also confirmed the presence of the rhamnose moiety in **6**. The downfield shift of C-6'' (δ 65.9) carbon resonance of the glucose unit was suggestive of the linkage of the rhamnose unit at the C-6'' position. In the ^1H NMR spectrum of **6**, the resonances at δ 7.08 (2H, d, $J = 9$ Hz) and 7.97 (2H, d, $J = 9$ Hz), which appeared as an AA'BB' system, were assigned to the H-3'/H-5' and H-2'/H-6' protons of the B ring, respectively. Moreover, 2 aromatic protons at δ 6.80 (1H, s) and 6.81 (1H, s) were readily attributed to H-3 and H-8, respectively. On the other hand, 2 methoxyl resonances at δ 3.82

(3H, s) and 3.88 (3H, s) were assigned to the C-4' and C-6 positions of the aglycon, respectively. Moreover, a ^1H NMR singlet resonance at δ 1.93 ppm suggested the presence of an acetyl group, attached to the C-4''' hydroxyl group of the rhamnose moiety, due to the downfield shifts of the H-4''' (δ 4.69, t, $J = 9.8$ Hz) and C-4''' (δ 74.3) resonances of rhamnose.

Based on the above NMR data, the structure of **6** was established as 4'''-O-acetyl pectolarin (linariin) 4,25,26.

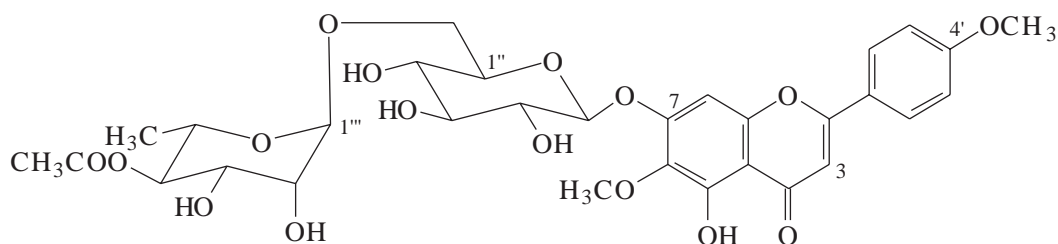


Figure 3. Linariin (**6**).

References

1. P.H. Davis, "*Linaria*" in Flora of Turkey and the East Aegean Islands Vol. 6, ed. P.H. Davis, pp. 654-672, University Press, Edinburgh, 1988.
2. T. Baytop, "Therapy with Medicinal Plants (Past and Present)", 2nd ed. p. 373, Nobel Tıp Kitabevleri Ltd., İstanbul, 1999.
3. A. San Feliciano, M. Gordaliza, J.M.M. Del Corral and M.L.De La Puente, **Phytochemistry**, **33**, 631-633 (1993).
4. H. Otsuka, **J. Nat. Prod.**, **55**, 1252-1255 (1992).
5. L.P. Smirnova, G.G. Zapesochnaya, V.I. Sheichenko and A.I. Bankovskii, **Khim. Prir. Soedin.**, **3**, 313-318 (1974).
6. M.P. Yuldashev, E.K. Batirov and V.M. Malikov, **Khim. Prir. Soedin.**, **1**, 38-41 (1996).
7. A. Bianco, M. Guiso, R.A. Mazzei, F. Piccioni, M. Nicoletti, M. Serafini and M. Ballero, **Fitoterapia**, **67**, 364-366 (1996).
8. N.V. Handjieva, E.I. Ilieva, S.L. Spassov and S.S. Popov, **Tetrahedron**, **49**, 9261-9266 (1993).
9. I. Kitagawa, T. Tani, K. Akita and I. Yosioka, **Chem. Pharm. Bull.**, **21**, 1978-1987 (1973).
10. H. Otsuka, **Phytochemistry**, **37**, 461-465 (1994).
11. I. Laakso, P. Virkajarvi, H. Airaksinen and E. Varis, **J. Chromatogr.**, **505**, 424-428 (1990).
12. H. Otsuka, **Phytochemistry**, **32**, 979-981 (1993).
13. S. Öksüz and G. Topçu, **Phytochemistry**, **26**, 3082-3084 (1987).
14. S. Ita, M. Kodama, M. Sunagawa, and T. Oba, **Tetrahedron Lett.**, **34**, 2905-2908 (1969).
15. J.W. Blunt and M.H.G. Munro, **Org. Mag. Resonance**, **13**, 26-27 (1980).

16. S. Xu, G.W. Patterson and K. Schmid, **Phytochemistry**, **25**, 1883-1886 (1986).
17. N.V. Kovganko and Z.N. Kashkan, **Khim. Prir. Soedin.**, **5**, 541-565 (1999).
18. R.A. Moreau, B.D. Whitaker and K.B. Hicks, **Progress in Lipid Research**, **41**, 457-500 (2002).
19. H. Kojima, N. Sato, A. Hatano and H. Ogura, **Phytochemistry**, **29**, 2351-2355 (1990).
20. B.V. Charlwood and D.V. Banthorpe, "Terpenoids", Vol. 7, eds. P.M. Dey and J.B. Harborne, Methods in Plant Biochemistry, pp. 369-433, Academic Press Ltd., London, 1991.
21. V. Piironen, D.G. Lindsay, T.A. Miettinen, J. Toivo and A.M. Lampi, **J. Sci. Food Agric.**, **80**, 939-966 (2000).
22. L.J. El-Naggar and J.L. Beal, **J. Nat. Prod.**, **43**, 649-776 (1980).
23. O. Sticher, **Phytochemistry**, **10**, 1974-1975 (1971).
24. J.L. Marco, **Phytochemistry**, **24**, 1609-1610 (1985).
25. N. Morita, M. Shimizu, M. Arisawa and K. Kobayashi, **Yakugaku Zasshi**, **94**, 913-916 (1974).
26. J.B. Harborne and T.J. Mabry, "The Flavonoids: Advances in Research", pp. 69, Chapman and Hall Ltd., London, 1982.