# Characterization of diterpenoids in the bark of *Pseudolarix kaempferi* by HPLC-ESI/MS<sup>n</sup>

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**Abstract**: Fragmentation behavior of diterpenoids was investigated by ESI/MS<sup>*n*</sup> and the qualitative analysis of diterpenoids in the bark of *Pseudolarix kaempferi* was performed using high-performance liquid chromatography/ multi-stage mass spectrometry (HPLC-ESI/MS<sup>*n*</sup>). The characteristic fragmentation behaviors of the diterpenoids are the cleavages of the lactone ring and C<sub>4</sub>-O bond. Furthermore, the eliminations of substituent groups at C-18, C-7 and C-8 can also be observed in the MS<sup>*n*</sup> (n = 3-4) spectra. For C-4 acetoxy substitued diterpenoids, [M+Na-60]<sup>+</sup> and [M-H-104]<sup>-</sup> are the base peaks of MS<sup>2</sup> spectra in the positive and negative ionization modes, respectively. For C-4 hydroxyl substitued diterpenoids, [M+Na-44]<sup>+</sup> and [M-H-62]<sup>-</sup> are the base peaks of MS<sup>2</sup> spectra in positive and negative ionization modes, [M+Na-44]<sup>+</sup> is the base peak of MS<sup>2</sup> spectra in positive ionization mode. These fragmentation rules were successfully exploited in the identification of diterpenoids in methanol/water (6 : 4) extract of *P. kaempferi* by LC-MS in positive ionization mode. A total of 9 diterpenoids were identified or tentatively characterized, and one of them is reported here for the first time. The described method could be utilized for the sensitive and rapid qualitative analysis of *P. kaempferi*.

**Key words**: *Pseudolarix kaempferi*; diterpenoid; quality control; qualitative analysis; HPLC-ESI/MS<sup>*n*</sup>; fragmentation

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# 土荆皮中二萜类化合物的 HPLC-ESI/MS" 鉴定

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摘要:使用高效液相色谱-多级质谱法(HPLC-ESI/MS")研究了土荆皮二萜的 ESI/MS"裂解行为,并对土荆皮中的二萜类化合物进行了定性分析。土荆皮二萜的特征裂解是内酯环和 C<sub>4</sub>-O 键的断裂。此外, C-18、C-7 和 C-8 位取代基的离去在多级质谱(MS", *n* = 3~4)中也可观察到。对于 C-4 位乙酰氧基取代的二萜, [M+Na-60]<sup>+</sup>和[M-H-104]<sup>-</sup>分别是正、负离子模式下二级质谱的基峰。对于 C-4 位羟基取代的二萜, [M+Na-44]<sup>+</sup>和[M-H-62]<sup>-</sup>分别是正、负离子模式下二级质谱的基峰。对于 C-18 位葡萄糖苷化或酯化的二萜, [M+Na-44]<sup>+</sup>是正离子模式下

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二级质谱的基峰。利用上述裂解规律,使用 LC-MS 方法在正离子模式下鉴定了土荆皮甲醇-水 (6:4) 提取物中的二萜类化合物,共有 9 个二萜被鉴定或初步推测,其中一个是首次报道的化合物。所述方法可用于土荆皮的灵敏快速定性分析。

关键词: 土荆皮; 二萜; 质量控制; 定性分析; 高效液相色谱-多级质谱; 裂解

Tu-Jing-Pi, the root and trunk bark of *Pseudolarix kaempferi* Gord. (Pinaceae), is commonly used as a traditional Chinese medicine for the treatment of skin diseases caused by fungal infections. Pharmacological research revealed that it had significant antifungal<sup>[1]</sup>, antitumor<sup>[2]</sup> and antifertility activities<sup>[3]</sup>, as well as potent sclerosing effect on gallbladder in rabbit<sup>[4]</sup>. Since 1980s, systematic phytochemical studies on the bark, seeds and leaves of *P. kaempferi* have led to the isolation of more than seventy compounds, including diterpenoids<sup>[1, 2]</sup>, triterpenoids<sup>[5]</sup>, triterpene lactones<sup>[6]</sup> and phenolic compounds<sup>[7]</sup>. The characteristic diterpenoids obtained from the bark were reported to be responsible for the biological activities of this drug<sup>[1-3]</sup>.

Therefore, the quality control of P. kaempferi should be focused on the analyses of the diterpenoids, which are of great importance for the quality of this drug. In our previous researches, a quantitative HPLC-DAD method has been established for simultaneous determination of seven major diterpenoids in *P. kaempfer*<sup>[8]</sup>. Furthermore, qualitative analysis of the drug is also necessary to ensure its authenticity and efficiency. However, HPLC-UV method is sometimes incompetent for the qualitative analysis because of the limitation of available reference standards necessary for the identification and the limited structural information provided by UV detector. High-performance liquid chromatography/multi-stage mass spectrometry (HPLC-MS<sup>n</sup>) technique combines the high separation ability of HPLC and the structural characterization function of MS and can rapidly assign the peaks without reference standards. Moreover, the sensitive and selective MS detector allows detection and identification of minor or even trace constituents from a micro-scale sample. Therefore, it is becoming an ever-increasing popular tool in the rapid characterization of chemical constituents in plant extracts<sup>[9, 10]</sup>.

The fragmentation mechanisms of diterpenoids isolated from *P. kaempferi* in electron impact (EI) source have been reported in  $1984^{[11]}$ . Subsequently, no extensive investigation has been performed on their mass spectra. Electrospray ionization (ESI) is one of the most frequently used ion sources in HPLC-MS<sup>*n*</sup>

analysis. In this paper, the ESI/MS fragmentation behavior of diterpenoids was studied for the first time, and was further applied to the identification of diterpenoids in a methanol/water (6:4) extract of *P. kaempferi*. Eight diterpenoids were identified, and one new component is tentatively characterized. The established method was valuable and dependable for the sensitive and rapid identification of diterpenoids in *P. kaempferi*.

## Materials and methods

Chemicals and reagents Pseudolaric acid A, pseudolaric acid B, pseudolaric acid C, pseudolaric acid C<sub>2</sub>, deacetylpseudolaric acid A, 11S-deacetylpseudolaric acid A, demethoxydeacetoxypseudolaric acid B, pseudolaric acid F, pseudolaric acid G, pseudolaric acid A  $O-\beta$ -D-glucopyranoside, pseudolaric acid B  $O-\beta$ -Dglucopyranoside, deacetylpseudolaric acid A O-β-Dglucopyranoside, deacetylpseudolaric acid A 2, 3dihydroxypropyl ester, and deacetylpseudolaric acid B 2, 3-dihydroxypropyl ester (Figure 1) were isolated from the bark of P. kaempferi and identified by their NMR and MS data. Their purities were not less than 95% by HPLC analysis. The bark of P. kaempferi was identified by Dr. GUO Hongzhu and a voucher specimen (040309TJP01) was deposited in the Department of Natural Medicines, School of Pharmaceutical Sciences, Peking University Health Science Center, China.

HPLC grade methanol, analytical grade acetic acid and other solvents used for compound isolation were all purchased from Beijing Chemical Engineering Factory (Beijing, China). The deionized water was prepared from Millipore water purification system (Millipore, Milford, MA, USA) and was filtered with 0.45  $\mu$ m membranes.

**Sample preparation** The dried powders of *P. kaempferi* samples (0.2 g, 60 mesh) were accurately weighed and soaked in 10 mL of methanol/water (6 : 4) solution at room temperature for 0.5 h then extracted at 80 °C for 0.5 h. The resultant mixture was adjusted to the original weight and filtered through filter paper and then 0.45  $\mu$ m membrane. Aliquots of 10  $\mu$ L were injected for LC-MS analysis.



14 Deacetylpseudolaric acid B 2, 3- H CH<sub>2</sub>CH(OH)CH<sub>2</sub>OH COOMe dihydroxypropyl ester (DPBP)







Figure 1 Structures of 14 diterpenoids

**HPLC conditions** The analyses were performed using an Agilent 1100 liquid chromatography system (Agilent, Waldbronn, Germany) equipped with a quaternary pump, a diode-array detector (DAD), an autosampler, and a column compartment. The samples were separated on an Inertsil ODS-3 column (250 mm × 4.6 mm, 5  $\mu$ m). The mobile phase consisted of methanol and water containing 0.3% acetic acid. A gradient program was used as follows: a linear gradient from 55% methanol to 90% methanol in 40 min. The mobile phase flow rate was 0.6 mL·min<sup>-1</sup>, and column temperature was set at 40  $^{\circ}$ C. The DAD recorded UV spectra in the range from 190 – 400 nm, and the HPLC chromatogram was monitored at 262 nm.

Mass spectrometry For HPLC/MS analysis, a Finnigan LCQ Advantage ion trap mass spectrometer (Thermo Finnigan, San Jose, CA, USA) was connected to the Agilent 1100 HPLC system via an ESI interface. The LC effluent was introduced into the ESI source in a post-column splitting ratio of 3:1. Ultra-high purity helium (He) was used as the collision gas and high purity nitrogen (N<sub>2</sub>) as the nebulizing gas. Pure standards were dissolved in MeOH- $H_2O(6:4)$  to obtain solutions containing 0.1  $mg \cdot mL^{-1}$  and infused into the ESI source using a syringe pump at a flow rate of 2.5  $\mu$ L·min<sup>-1</sup>. The operating parameters were optimized for each standard. The operating parameters in the positive ionization mode for plant extracts were as follows: ion spray voltage, 4.5 kV; sheath gas (N<sub>2</sub>) pressure, 50 arbitrary units; auxiliary gas (N<sub>2</sub>) pressure, 10 units; capillary temperature, 300 °C; capillary voltage, 30 V. For full-scan MS analysis, the spectra were recorded in the range of m/z 100 – 800. A data-dependent program was used in the liquid chromatography/tandem mass spectrometry analysis so that the two most abundant ions in each scan were selected and subjected to MS<sup>n</sup> (n=2-4) analyses. The collision-induced dissociation (CID) energy was adjusted to 50%. The isolation width of the precursor ions was 1.5 Th.

#### **Results and discussion**

#### **1** Tandem mass spectrometry of pure standards

The ionization of diterpenoids was performed in both positive and negative ionization modes and the responses were good in the both of the two modes. So the fragmentation behaviors of diterpenoids in the both of the two modes were investigated and compared.

The 14 reference compounds were introduced into the ESI source by continuous infusion. The base peak ions were subjected to  $MS^2$  fragmentation and the prominent ions in each  $MS^2$  spectrum were then selected for further  $MS^n$  analysis (n = 3 - 4). Normalization collision energy was in the range of 20%–50%. Except for the lactone ring, the skeleton of the diterpenoids is stable in the ESI source. Hence the substitution groups determine the fragmentation behaviors of the diterpenoids. The fragmentation patterns of the diterpenoids could be classified into four groups according to their substitutions at C-4 and C-18 positions. The fragmentation pathways are shown in Scheme 1.



Scheme 1 Major fragmentation pathways of diterpenoids in positive and negative ionization modes

Group I: C-4 acetoxy and C-18 carboxyl 1.1 Compounds in this group have acetoxy and carboxyl substituted at C-4 and C-18, respectively, including PA, PB, PC<sub>2</sub>, PF and PG. In positive ionization mode, the full scan mass spectrum of PB gave a  $[M+Na]^+$  ion at m/z 455 and this ion was selected for MS<sup>n</sup> (n = 2 - 4) analysis. A base peak ion at m/z 395 ([M+Na-60]<sup>+</sup>) was observed in the MS<sup>2</sup> spectrum, resulting from the loss of CH<sub>3</sub>COOH. The ions at m/z 411 ([M+Na-44]<sup>+</sup>) and m/z 351 ([M+Na-104]<sup>+</sup>) were attributed to elimination of CO2 and concurrent losses of CH3COOH and CO<sub>2</sub>, respectively. The ion at m/z 395 was subjected to  $MS^3$  analysis and produced a  $[M+Na-60-44]^+$  ion at m/z 351, which underwent loss of CO<sub>2</sub> and CH<sub>3</sub>OH to generate ions at m/z 307 and m/z 319 in MS<sup>4</sup> spectrum, respectively. The loss of 32 Da (CH<sub>3</sub>OH) provided characteristic fragmentation information for compounds with C-7 methyl formate substitution. The results obtained in this study led to a proposed fragmentation pathway of PB in positive ionization mode as illustrated in Figure 2. PB could lose two molecules of  $CO_2$ , owing to its structural moieties of lactone ring and C-18 carboxyl group. It was assumed that PB underwent previous loss of lactone ring and following loss of C-18 carboxyl because the cleavage of lactone ring generated an ion with very stable conjugated structure, i.e. the ion at m/z 351 in Figure 2.

The other compounds in this group, viz. PA, PC<sub>2</sub>, PF and PG, also showed prominent ions of  $[M+Na-60]^+$ , as well as the minor ions of  $[M+Na-104]^+$  in their MS<sup>2</sup> spectra. The product ions could further lost CO<sub>2</sub> and the substituent groups at C-8, showing characteristic fragmentation information corresponding to their structures. For example, the loss of C-8 carbonyl and hydroxyl group resulted in the ions of m/z 293 (425 $\rightarrow$  321 $\rightarrow$  293) and m/z 305 (427 $\rightarrow$  323 $\rightarrow$  305) in MS<sup>3</sup> of PF and PG, respectively.

In negative ionization mode, compounds in this group usually gave  $[M-H]^-$  ions, which were subjected to further fragmentation and yielded  $[M-H-104]^-$  ions as base peaks, as well as the  $[M-H-60]^-$  ions. The subsequent losses of COO<sup>-</sup> could also be observed in MS<sup>3</sup> spectra of  $[M-H-104]^-$  ions. Moreover, the loss of the C-7 substituent group (methyl free radical) was



Figure 2 The proposed fragmentation pathway of PB in positive ionization mode

not found in positive ionization mode.

**1.2 Group II: C-4 hydroxyl and C-18 carboxyl** Compounds in this group contain C-4 hydroxyl and C-18 carboxyl, including PC, DDPB, DPA and 11*S*-DPA. In positive ionization mode, fragmentation of the  $[M+Na]^+$  ion of PC at m/z 413 involved initial loss of CO<sub>2</sub> to generate the  $[M+Na-44]^+$  ion  $(m/z \ 369)$ . The ion at  $m/z \ 369$  was then subjected to MS<sup>3</sup> analysis and produced a prominent ion  $([M+Na-44-18]^+)$  at  $m/z \ 351$ , which then yielded an ion at  $m/z \ 319$  in the MS<sup>4</sup> spectrum, through loss of a molecule of CH<sub>3</sub>OH. Another minor ion at  $m/z \ 325$  was also observed in the MS<sup>3</sup> spectrum of the ion at  $m/z \ 369$ , resulting from the loss of CO<sub>2</sub>.

DDPB had almost the same fragmentation pathway as PC, except for the loss of CH<sub>3</sub>OH. DPA and 11*S*-DPA are configuration isomers and their fragmentation behaviors are identical. Fragmentation of  $[M+Na]^+$  ion at m/z 369 was triggered by initial loss of CO<sub>2</sub> (369 $\rightarrow$ 325), followed by simultaneous loss of CO<sub>2</sub> and CH<sub>3</sub> (325 $\rightarrow$ 266).

In negative ionization mode, compounds in this group usually showed  $[M-H]^-$  ions, which were selected for further MS<sup>2</sup> analyses and generated  $[M-H-62]^-$  ions as base peaks, as well as  $[M-H-44]^-$  ions. The  $[M-H-62]^-$  ions were originated from concurrent loss of COO<sup>-</sup> and H<sub>2</sub>O from  $[M-H]^-$  ion, and the subsequent losses of COO<sup>-</sup>, CH<sub>3</sub>OH or CH<sub>3</sub> could also be observed in MS<sup>3</sup> spectra corresponding to their substitutions at C-18 and C-7.

**1.3 Group III: C-18 carboxylic glucoside** The C-18 carboxyl of this group of compound is glucosylated,

such as PAG, PBG and DPAG. In positive ionization mode, the  $[M+Na]^+$  ion at m/z 573 of PAG underwent successive loss of 44 Da (CO<sub>2</sub>) and then 162 Da (glucosyl). The m/z 367 ion obtained subsequently produced ions at m/z 323 and m/z 307 via the same pathways as PA. PBG and DPAG had almost the same fragmentation pattern as PAG.

In negative ionization mode, compounds in this group usually showed remarkable  $[M+2H_2O-H]^-$  ions, which were selected for further  $MS^2$  analyses and ions formed via the loss of 198 Da ( $[M+2H_2O-H-36-162]^-$ ) were observed as base peaks and their further fragmentation behaviors were the same as the corresponding aglycones.

**1.4 Group IV: C-18 carboxylic ester** This group of compounds has a 2, 3-dihydroxypropyl group attached to C-18, forming an ester of the acid, such as DPAP and DPBP. The MS<sup>2</sup> spectrum of DPAP from the  $[M+Na]^+$  ion at m/z 443 in positive ionization mode gave a  $[M+Na-44]^+$  ion at m/z 399. The MS<sup>3</sup> spectrum of m/z 399 yielded m/z 381 and m/z 325, corresponding to loss of a molecule of H<sub>2</sub>O and a 2, 3-dihydroxypropyl group, respectively. DPBP exhibited the same fragmentation pathway as DPAP.

In negative ionization mode, the  $[M-H]^-$  ions of DPAP and DPBP were observed.  $[M-H-74]^-$  and  $[M-H-136]^-$  ions in the MS<sup>2</sup> spectra were observed owing to elimination of 2, 3-dihydroxypropyl group and concurrent losses of COO<sup>-</sup>, H<sub>2</sub>O and 2, 3-dihydroxypropyl group, respectively.

Except for the lactone ring, the skeleton of the diterpenoids is stable in the ESI source in positive and

negative ionization modes, therefore the substitution groups determine the fragmentation behaviors of the diterpenoids. The typical fragmentation behaviors of the diterpenoids are the cleavages of the lactone ring and  $C_4$ -O bond. Furthermore, the eliminations of substituent groups at C-18, C-7 and C-8 can also be observed in the MS<sup>3</sup> and MS<sup>4</sup> spectra. The major difference of the fragmentation behaviors between these two ionization modes is the base peak in the  $MS^2$ spectrum. In positive ionization mode, [M+Na-60]<sup>+</sup> or  $[M+Na-44]^+$  ion is the base peak in the MS<sup>2</sup> spectrum, which is attributed to the loss of C-4 acetoxy or laction ring. While [M-H-104] or [M-H-62] ion is the base peak in negative ionization mode, corresponding to the concurrent loss of lactone ring and C-4 substitution, i.e. acetoxy or hydroxyl.

### 2 HPLC-ESI/MS<sup>n</sup> analysis of *P. kaempferi* extract

Positive ionization mode was used in the HPLC/ ESI-MS<sup>n</sup> analysis of *P. kaempferi* extract because the acetic acid in the mobile phase strongly inhibited the ionization of diterpenoids in negative ionization mode. The HPLC-UV and total ion current (TIC) profiles of the extract of *P. kaempferi* crude drug were shown in Figure 3. Nine diterpenoids were detected and characterized from the extract of *P. kaempferi* (Table 1). Eight of them were unambiguously identified by comparing their retention times, UV and mass spectra with reference standards. For the unknown peak, the structure was tentatively established based on its MS<sup>n</sup> analysis, according to the general fragmentation



Figure 3 The HPLC-UV chromatogram monitored at 262 nm and ESIMS total ion current profile of *P. kaempferi* 

rules summarized above as well as the biogenesis.

**2.1 Group I** Peak 8 (retention time ( $t_R$ ) = 28.53 min) gave [M+Na]<sup>+</sup> ions at m/z 469 and yielded product ion at m/z 409 as base peak, as well as minor ions at m/z 425 and 365 in MS<sup>2</sup> analysis. The ion at m/z 409 was assigned as the loss of CH<sub>3</sub>COOH from the precursor ion at m/z 469, indicating the presence of C-4 acetoxy, which further generated ion at m/z 365 owing to the cleavage of lactone ring. The ion at m/z 377 was observed in MS<sup>3</sup> spectrum of m/z 409 and was attributed

 Table 1
 Characterization of diterpenoids in *Pseudolarix kaempferi* by HPLC-ESI/MS<sup>n</sup> (m/z, % of relative abundance). \*New compound

 No.
 (m/z) (m/z) 

 No.
 (m/z) (m/z) 

No.	$t_{\rm R}/{\rm min}$	Assiged identity	UV $\lambda_{max}/nm$	$[\mathbf{M}+\mathbf{Na}]^+ m/z$	HPLC-ESI/MS <sup>n</sup> $m/z$ (% base peak)
1	12.87	DPBP	236	487	MS <sup>2</sup> [487]: 443 (100), 425 (10)
2	13.54	PBG	266	617	MS <sup>2</sup> [617]: 573 (60), 557 (35), 513 (10), 455 (100), 395 (35)
					MS <sup>3</sup> [617→455]: 413 (15), 395 (100), 351 (15)
					MS <sup>4</sup> [617→455→395]: 363 (100), 351 (65)
3	15.48	$PC_2$	262	441	MS <sup>2</sup> [441]: 397 (5), 381 (100), 337 (70)
					MS <sup>3</sup> [441→381]: 337 (100)
4	15.75	PC	262	413	MS <sup>2</sup> [413]: 369 (100), 351 (1)
					MS <sup>3</sup> [413→369]: 351 (100), 325 (25)
5	16.15	DPAP	242	443	MS <sup>2</sup> [443]: 381 (100), 327 (25)
6	20.08	PAG	266	573	MS <sup>2</sup> [573]: 529 (100), 367 (10)
					MS <sup>3</sup> [573→529]: 367 (100)
					$MS^4$ [573 $\rightarrow$ 529 $\rightarrow$ 367]: 307 (100)
7	22.02	PB	260	455	MS <sup>2</sup> [455]: 411 (20), 395 (100), 351 (60)
					MS <sup>3</sup> [455→395]: 377 (20), 363 (50), 351 (100)
8*	28.53	PI	252	469	MS <sup>2</sup> [469]: 425 (20), 409 (100), 365 (50)
					MS <sup>3</sup> [469→409]: 391 (5), 377 (15), 365 (100)
9	29.25	PA	260	411	MS <sup>2</sup> [411]: 367 (90), 351 (100), 307 (90)

to the emanation of  $CH_3OH$ , suggesting the existence of C-7 methyl formate. The molecular mass of Peak 8 was 44 Da higher than known compound PF, in conjunction with the C-7 methyl formate and biogenesis relationship, it was therefore tentatively characterized as a new compound pseudolaric acid I (PI) as showed in Figure 4.



Figure 4 Proposed structure of PI

**2.2** Group II Peak 4 ( $t_R = 15.75 \text{ min}$ ) gave [M+Na]<sup>+</sup> ions at m/z 413 and produced ion at m/z 369 as base peak, as well as minor ion at m/z 351 in MS<sup>2</sup> analysis. The ion at m/z 351 was assigned as the concurrent loss of CO<sub>2</sub> and H<sub>2</sub>O from the precursor ion at m/z 413, indicating the presence of C-4 hydroxyl. Hence, Peak 4 was identified as pseudolaric acid C (PC) by comparing with the reference standard.

**2.3 Group** III Peak 2 ( $t_R = 13.54 \text{ min}$ ) gave [M+Na]<sup>+</sup> ions at m/z 617 and its MS<sup>2</sup> spectra produced ions at m/z 573, 557 and 455, corresponding to losses of CO<sub>2</sub>, CH<sub>3</sub>COOH and glycosyl, respectively. The ion at m/z 455 was further subjected to MS<sup>3</sup> analysis and yielded ion at m/z 395 as base peak, indicating the presence of C-4 acetoxy. The ion at m/z 413 in MS<sup>3</sup> was assigned as the loss of CH<sub>3</sub>OH from m/z 455, suggesting the existence of C-7 methyl formate. The MS<sup>4</sup> spectrum of m/z 395 gave ions at m/z 363 and 351, which were attributed to the cleavages of CH<sub>3</sub>OH and CO<sub>2</sub>, respectively. Thus, Peak 2 was identified as pseudolaric acid B *O*- $\beta$ -*D* glucopyranoside (PBG) by comparing with the reference standard.

**2.4 Group IV** Peak 1 ( $t_R = 12.87 \text{ min}$ ) gave  $[M+Na]^+$ ions at m/z 487 and its MS<sup>2</sup> spectra produced ions at m/z 443 and 425, corresponding to loss of CO<sub>2</sub> and concurrent losses of CO<sub>2</sub> and H<sub>2</sub>O, respectively. Therefore, Peak 1 was identified as deacetylpseudolaric acid B 2, 3-dihydroxypropyl ester (DPBP) by comparing with the reference standard. The identification of DPAP and DPBP from the methanol/water extract of *P. kaempferi* unambiguously demonstrated that they are natural products originated from the plant, but not artefacts.

## Conclusions

In this study, the fragmentation behaviors of diterpenoids in the bark of P. kaempferi were investigated by using HPLC-ESI/MS<sup>n</sup> method. Except for the lactone ring, the skeleton of the diterpenoids is stable in the ESI source in the two ionization modes. The typical fragmentation behaviors of the diterpenoids are the cleavages of the lactone ring and C<sub>4</sub>-O bond. Furthermore, the eliminations of substituent groups at C-18, C-7 and C-8 can also be observed in the  $MS^n$  (n =3-4) spectra. The major difference of the fragmentation behaviors between these two ionization modes is the base peak in the  $MS^2$  spectrum. For C-4 acetoxy substituted diterpenoids, [M+Na-60]<sup>+</sup> and [M-H-104]<sup>-</sup> are the base peaks in the positive and negative ionization For C-4 hydroxyl substituted mode, respectively. diterpenoids,  $[M+Na-44]^+$  and  $[M-H-62]^-$  are the base peaks in the positive and negative ionization mode, For C-18 glucosylated or esterized respectively. diterpenoids,  $[M+Na-44]^+$  is the base peak in positive ionization mode.

These deduced rules were successfully exploited in the identification of diterpenoids in the bark of *P. kaempferi* by LC/MS in positive ionization mode. A total of 9 diterpenoids were identified or tentatively characterized, and one of them is reported here for the first time. The described method could be utilized for the sensitive and rapid qualitative analysis of *P. kaempferi* and its pharmaceutical preparations. This method is of great significance for the characterization of minor diterpenoids *in vivo* after dosing with *P. kaempferi*, so as to clarify their metabolic pathways and remedial mechanisms.

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・消息・

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