

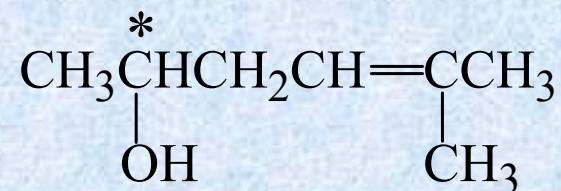
# 同步训练七

醇、酚、醚

## 一、选择题：

1. 5-甲基-4-己烯-2-醇的立体异构体数目有: ( A )

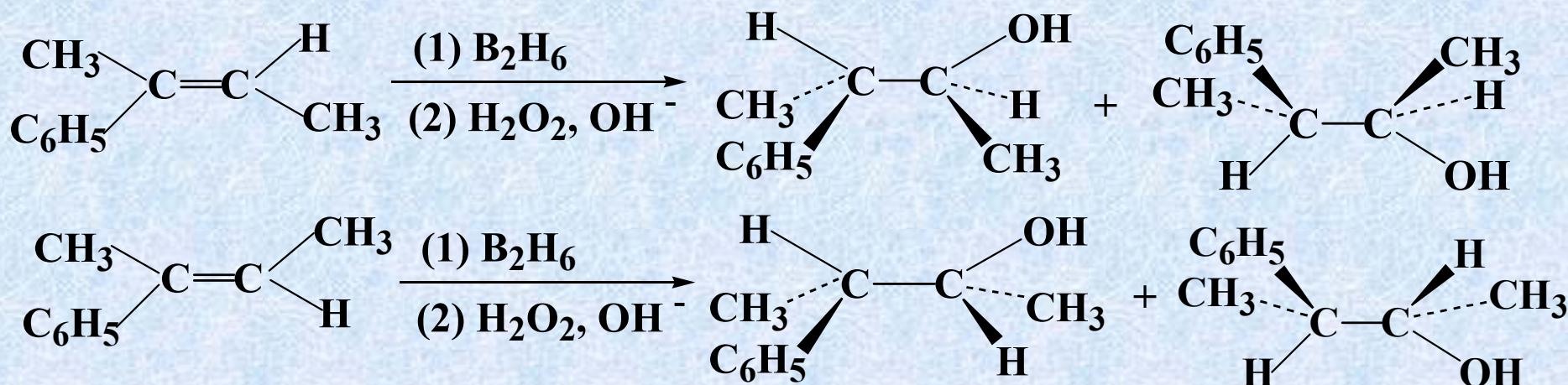
- A) 2      B) 4      C) 5      D) 6



5-甲基-4-己烯-2-醇

2. (Z)-2-苯基-2-丁烯与(E)-2-苯基-2-丁烯经硼氢化反应间接水合制得的醇中,下列哪种说法正确: ( D )

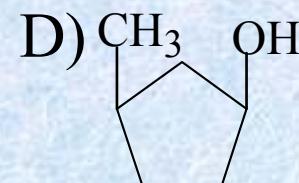
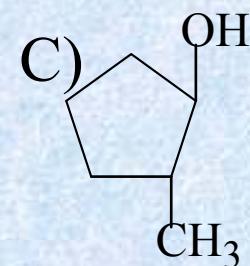
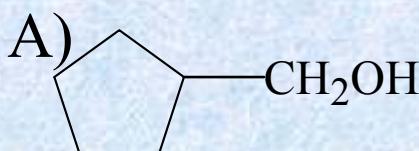
- A) 前者制得的醇为一对对映体,后者仅一个化合物
- B) 两者制得的醇均为一对顺反异构体
- C) 前者制得的醇为一个化合物,后者制得的醇为一对外消旋体
- D) 两者制得的醇均为一对外消旋体



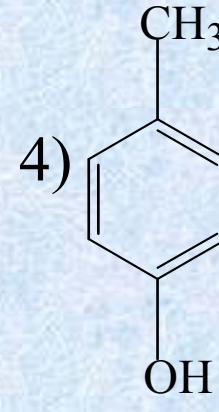
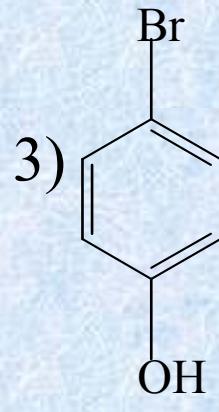
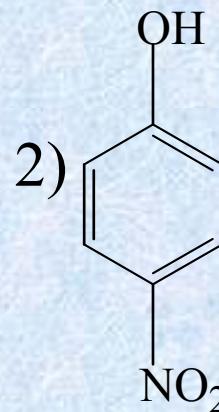
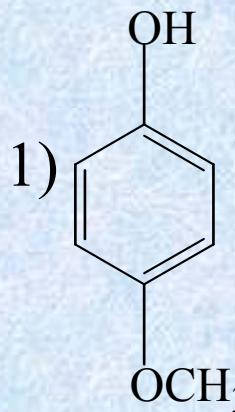
3. 环己醇与下列试剂不反应的是: ( B )

- A) 浓 $\text{H}_2\text{SO}_4$ ,  $\triangle$       B) 稀, 冷 $\text{KMnO}_4$   
C)  $\text{CH}_3\text{MgI}$ ,  $\text{Et}_2\text{O}$       D)  $\text{K}_2\text{Cr}_2\text{O}_7 + \text{H}_2\text{SO}_4$ ,  $\triangle$

4. 下列各醇与 $\text{HBr}$ 反应的相对速率最快的是:( B )



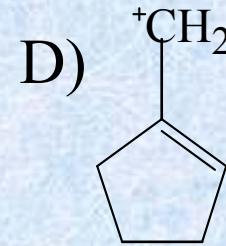
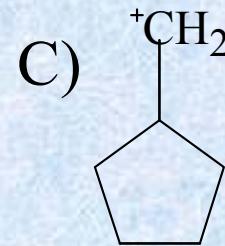
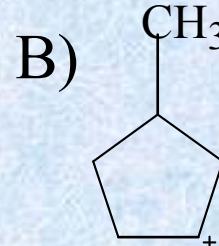
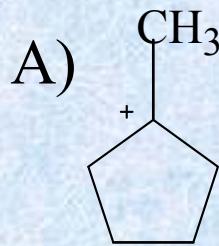
5. 下列各化合物的酸性大小:( A )



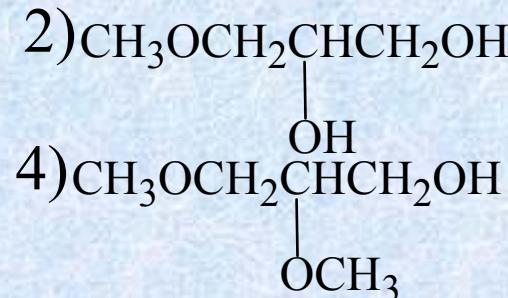
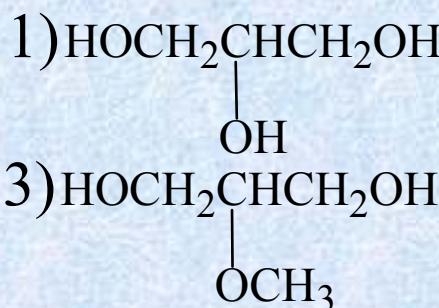
- A)  $2 > 3 > 4 > 1$   
C)  $4 > 2 > 1 > 3$

- B)  $2 > 4 > 1 > 3$   
D)  $3 > 2 > 1 > 4$

6. 下列各碳正离子最稳定的是: ( D )

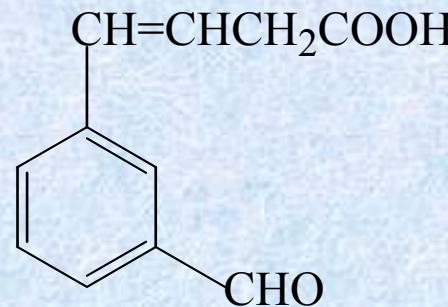


7. 下列醇沸点顺序为:( B )

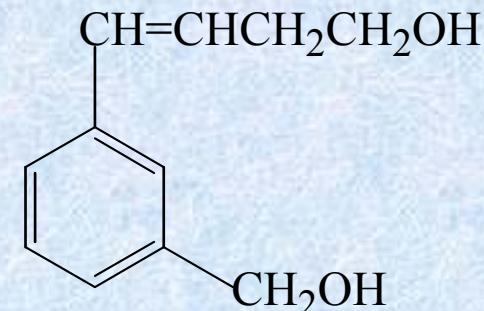


- A) 1>2>3>4 B) 1>3>2>4 C) 2>4>1>3 D) 3>2>1>4

8. 将化合物



还原为



的最好方法是:( B )

- A) NaBH<sub>4</sub>, EtOH  
C) H<sub>2</sub>, Pt/EtOH

- B)(1) LiAlH<sub>4</sub>, Et<sub>2</sub>O (2) H<sub>2</sub>O  
D) Na, EtOH

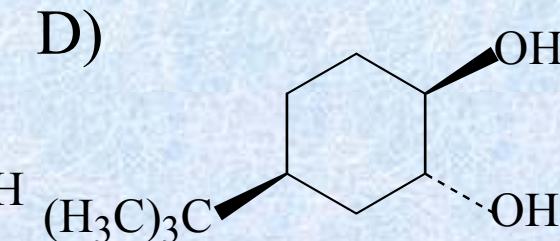
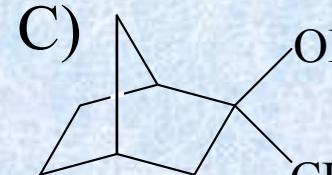
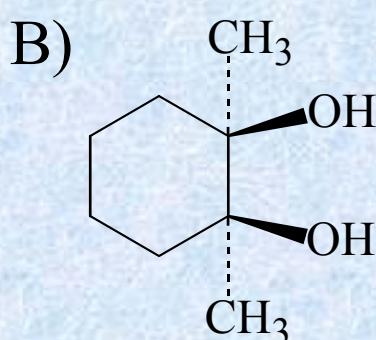
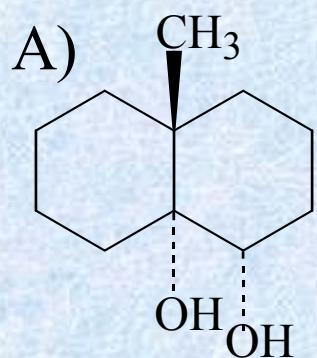
9. 下列碱性最大的是: ( C )



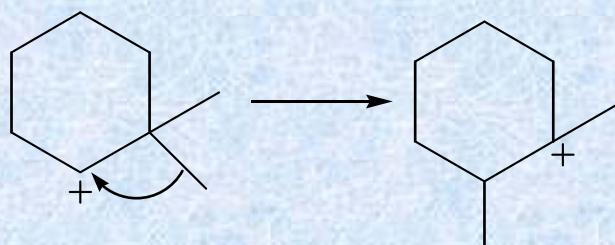
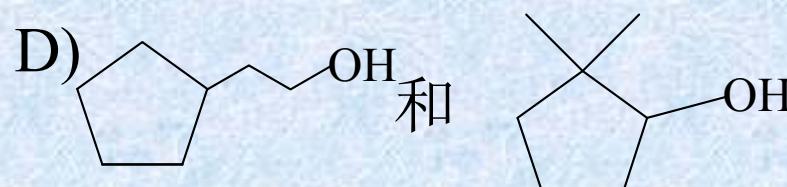
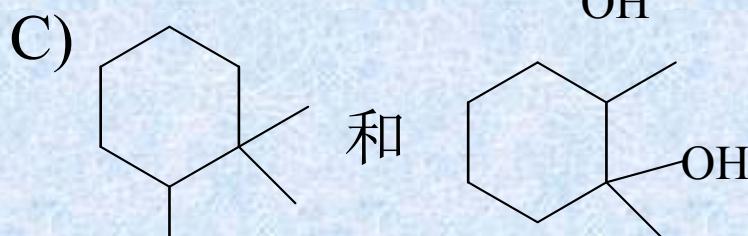
酸性:



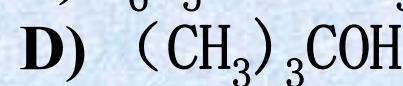
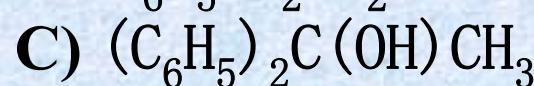
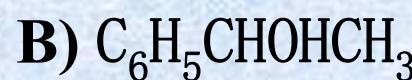
10. 下列各邻二醇, 不被 $\text{HI}\text{O}_4$  氧化的是: ( D )



11. 下面四组醇, 哪一组用酸处理后, 2个化合物成为相同的C<sup>+</sup> ( C )

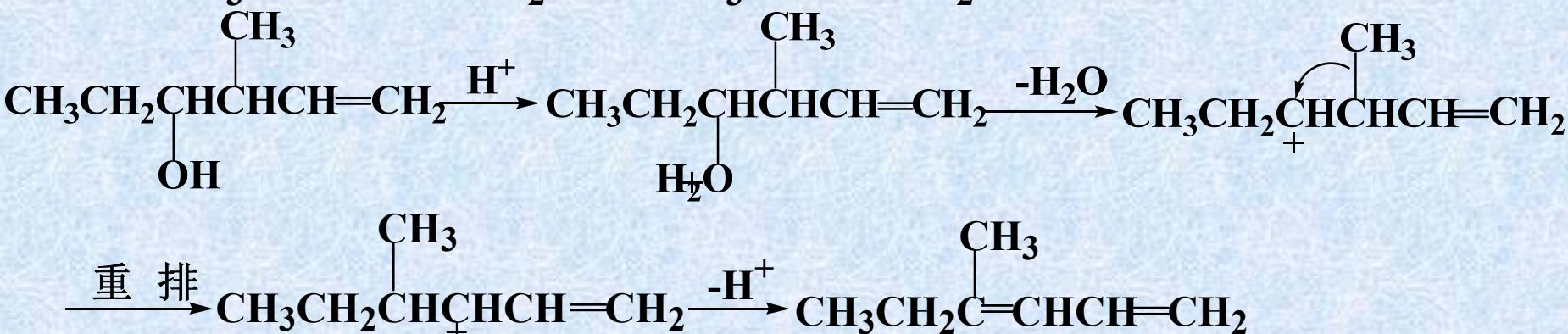


12. 下面4个醇, 最易脱水的是: ( C )



13. 下列4个化合物, 最易在酸催化下进行消除反应的是: ( C )

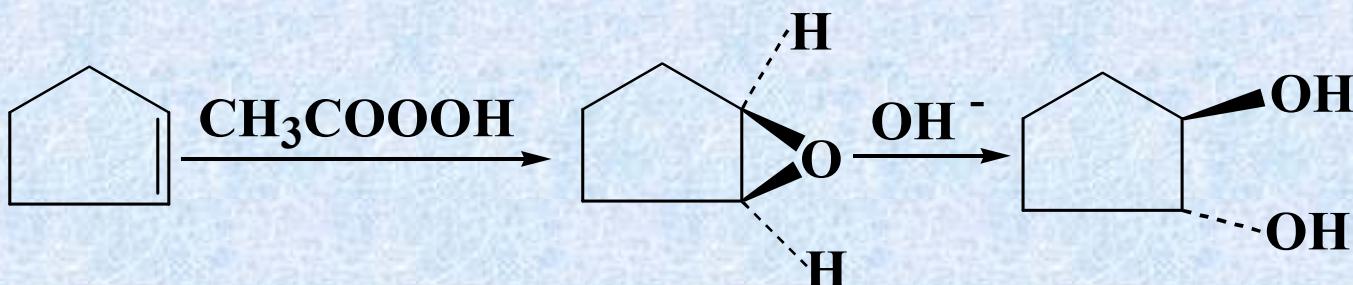
- A) HOCH<sub>2</sub>CH<sub>2</sub>CH=CH<sub>2</sub>    B) CH<sub>3</sub>CH<sub>2</sub>CH(OH)CH<sub>2</sub>CH=CH<sub>2</sub>  
C) CH<sub>3</sub>CH<sub>2</sub>CH(OH)CH(CH<sub>3</sub>)CH=CH<sub>2</sub>  
D) CH<sub>3</sub>CH(OH)CH<sub>2</sub>CH(CH<sub>3</sub>)CH=CH<sub>2</sub>



14. 由环戊烯转化为反-1,2-环戊二醇应采用的方法是: ( D )

- A) KMnO<sub>4</sub>, H<sub>2</sub>O, OH<sup>-</sup>  
C) (BH<sub>3</sub>)<sub>2</sub>, H<sub>2</sub>O, OH<sup>-</sup>

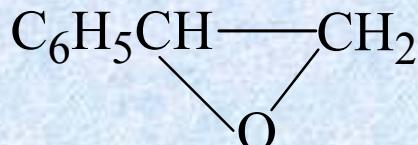
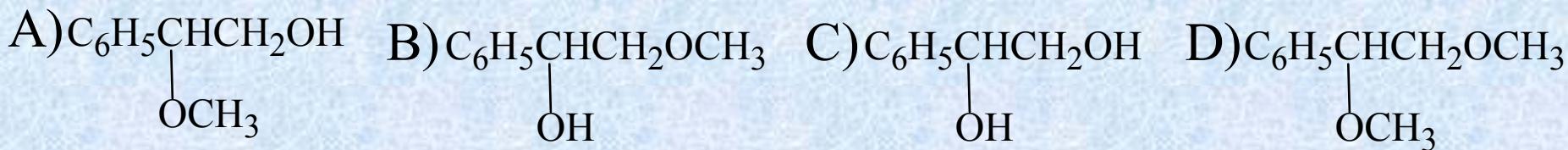
- B) O<sub>3</sub>, Zn+H<sub>2</sub>O  
D) (1)CH<sub>3</sub>COOOH (2)OH<sup>-</sup>



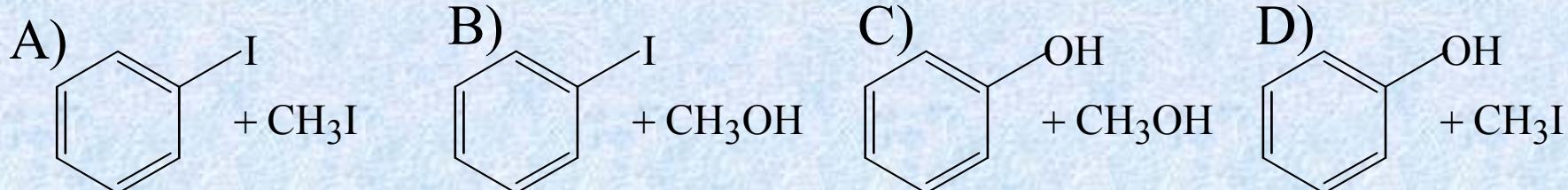
15. 用Williamson合成法合成 $\text{CH}_3\text{CH}_2\text{OC}(\text{CH}_3)_3$ ,下列方法中最合适的是: ( A )

- A)  $\text{CH}_3\text{CH}_2\text{Br} + \text{NaOC}(\text{CH}_3)_3$       B)  $\text{CH}_3\text{CH}_2\text{ONa} + \text{BrC}(\text{CH}_3)_3$   
C)  $\text{CH}_3\text{CH}_2\text{Br} + \text{NaOCH}_2\text{CH}(\text{CH}_3)_2$  D)  $\text{CH}_3\text{CH}_2\text{ONa} + \text{BrCH}_2\text{CH}(\text{CH}_3)_2$

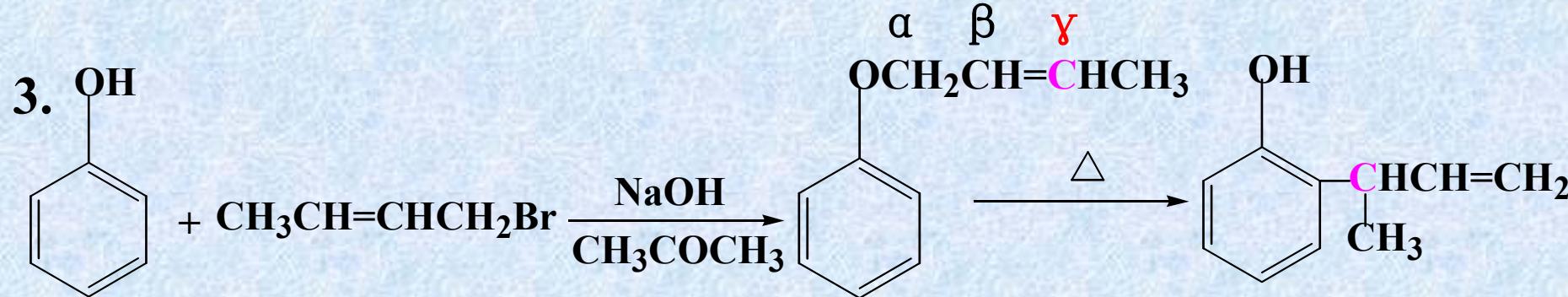
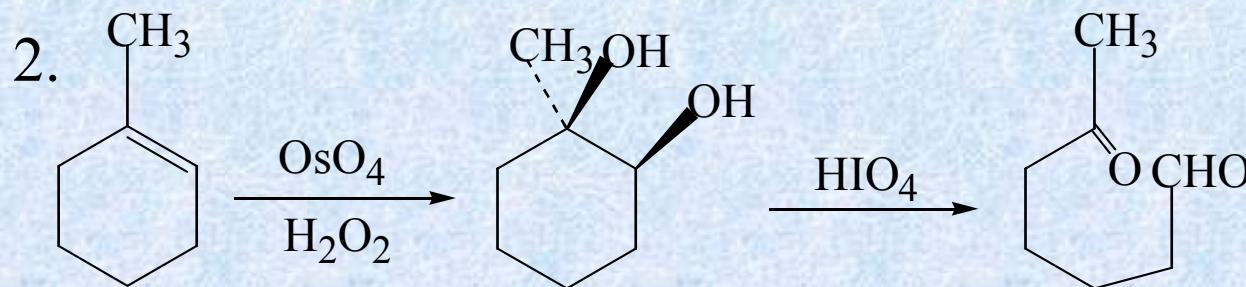
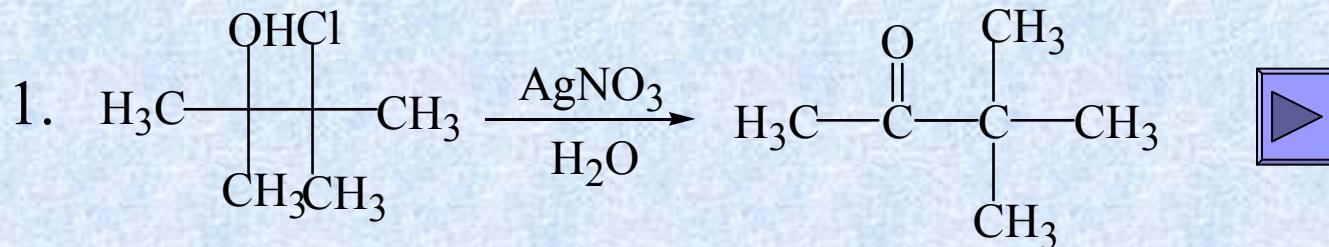
16. 氧化苯乙烯与 $\text{CH}_3\text{OH}$  和少量 $\text{CH}_3\text{ONa}$  反应的主要产物是( B )

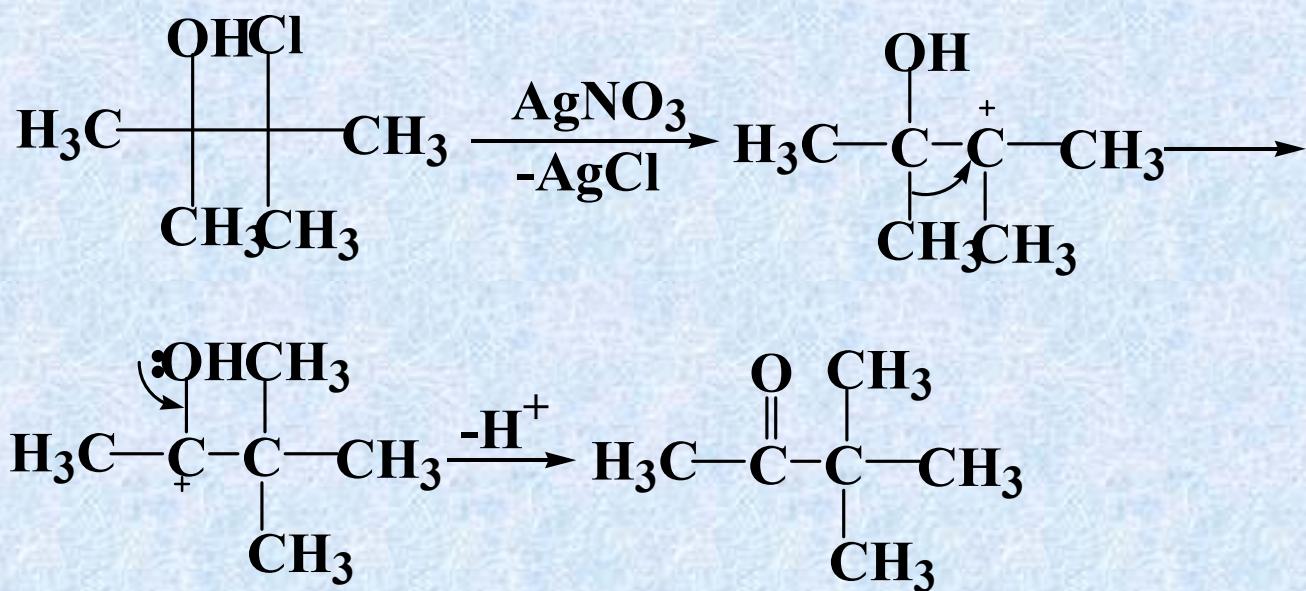


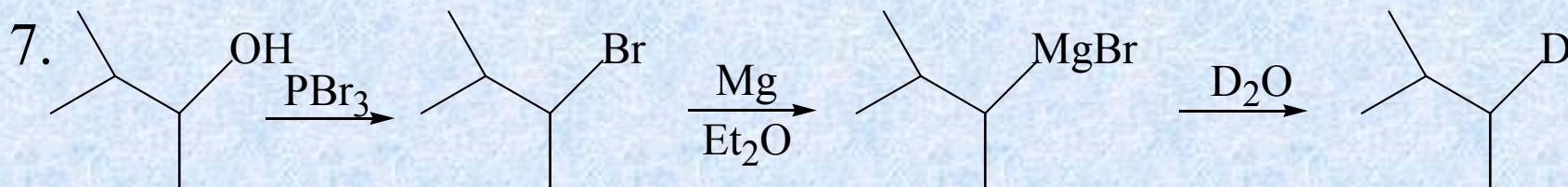
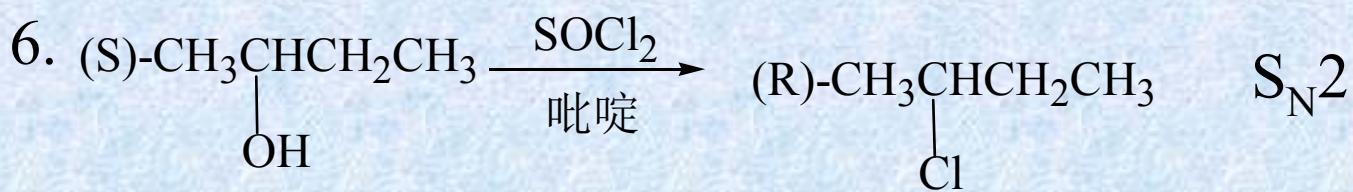
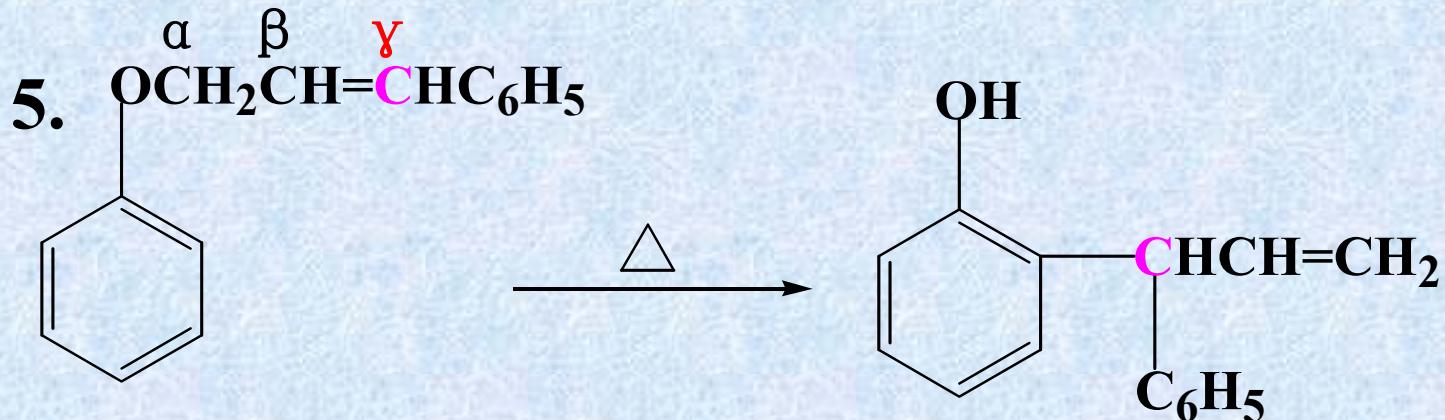
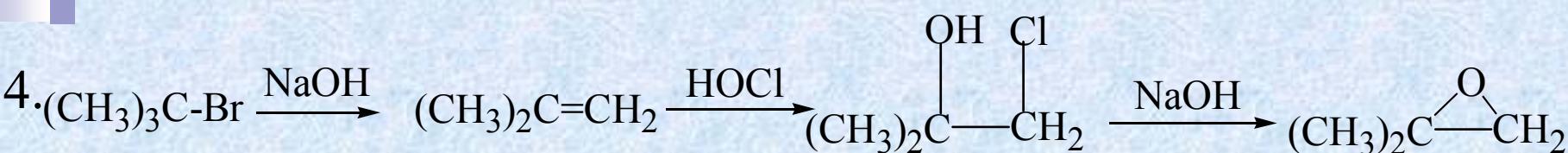
17.  $\text{PhOCH}_3$  由HI裂解的产物是: ( D )

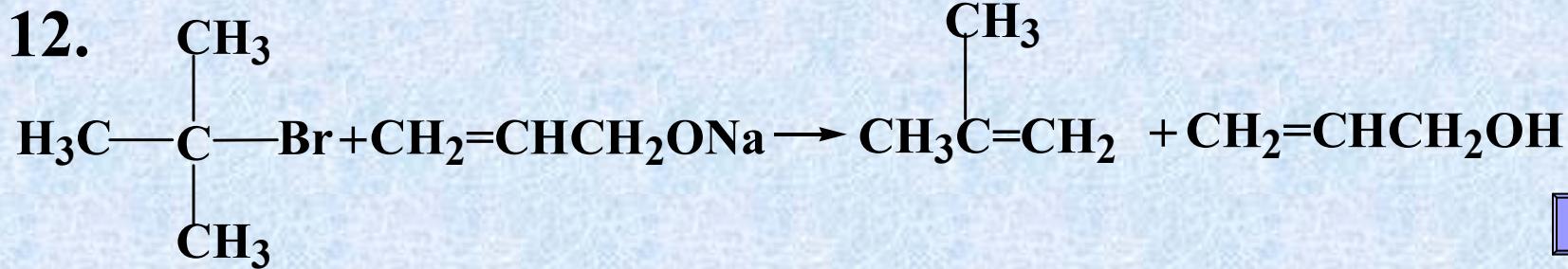
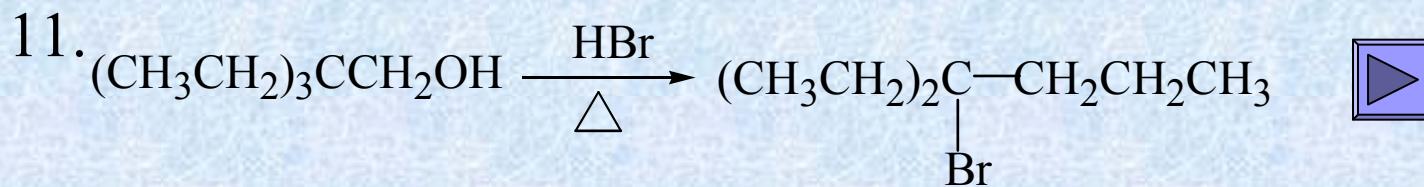
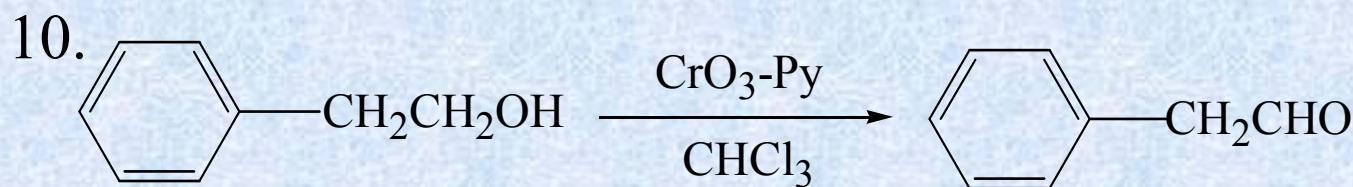
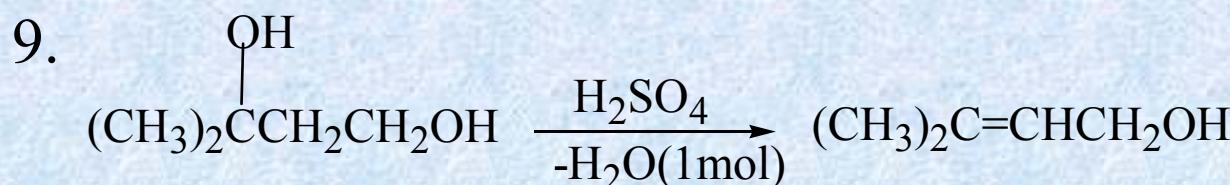
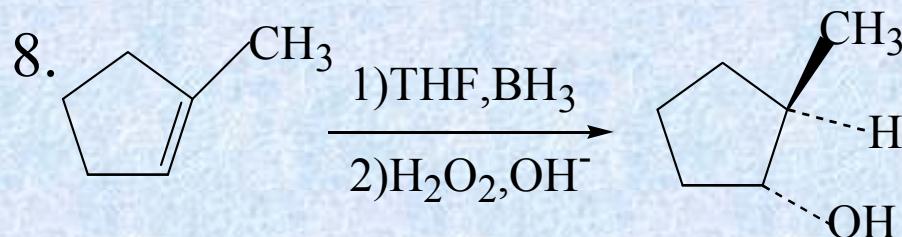


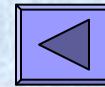
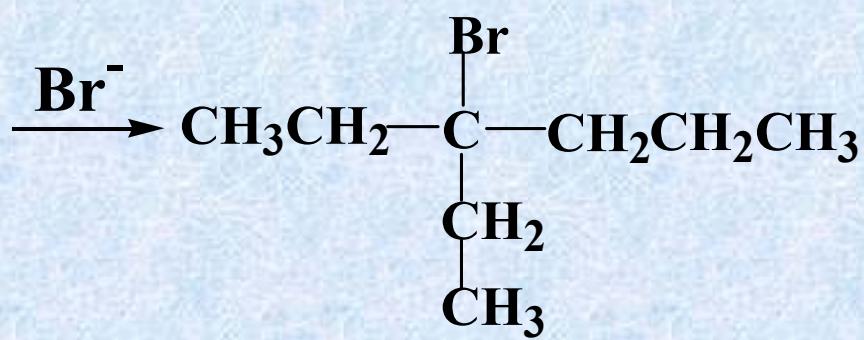
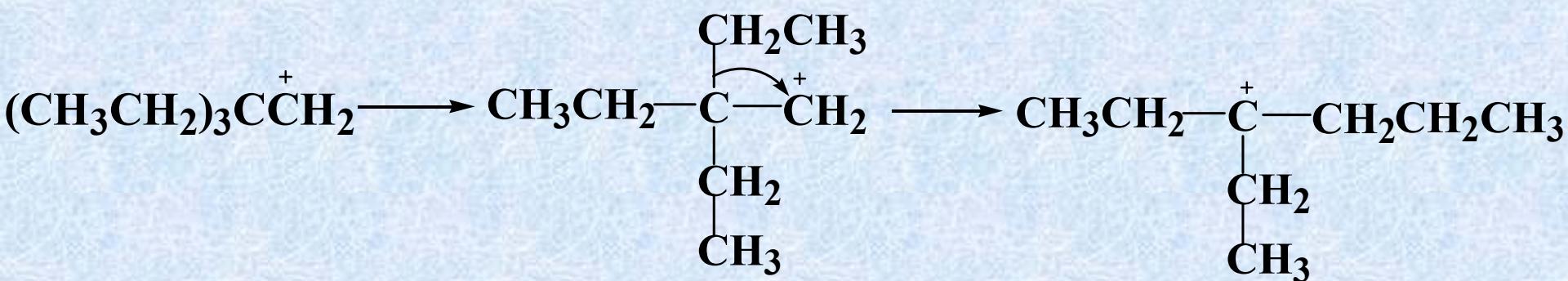
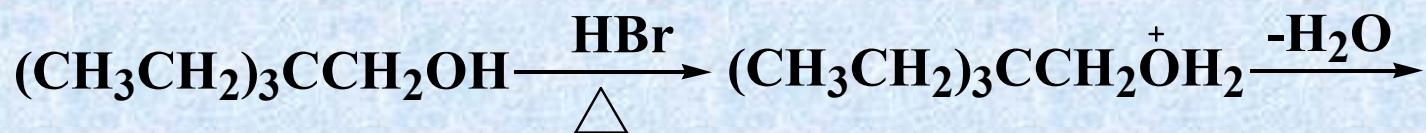
二、写出下列反应的主要有机产物或所需之原料、试剂(如有立体化学问题,请注明):



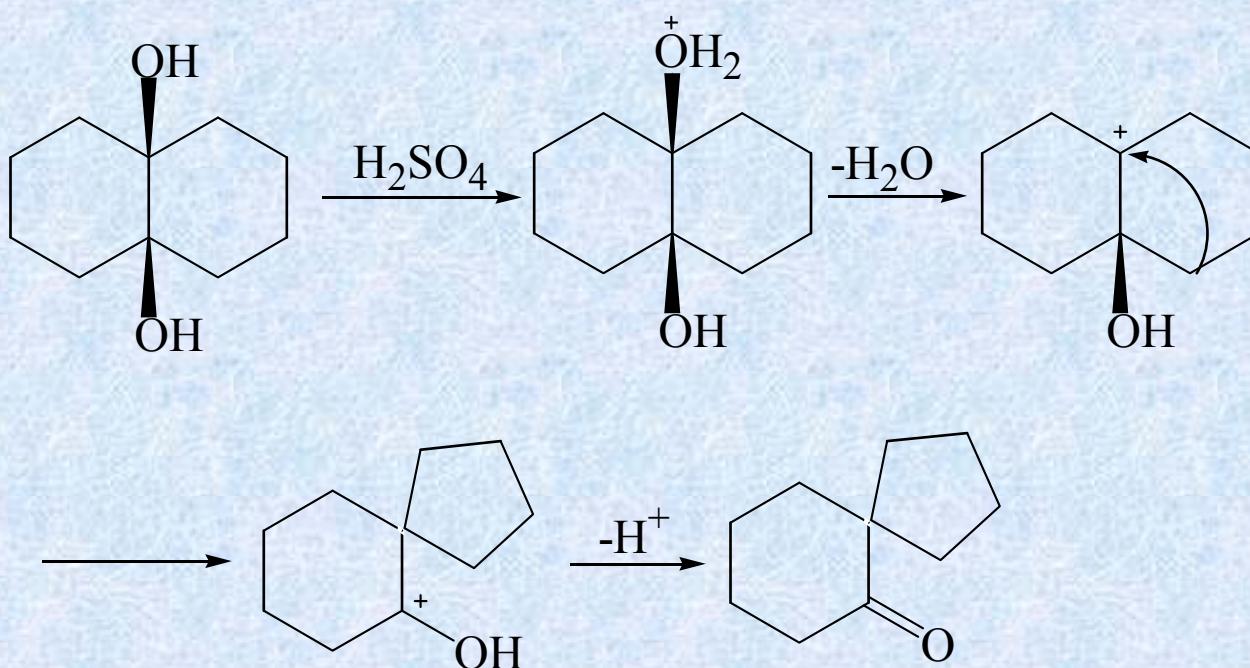
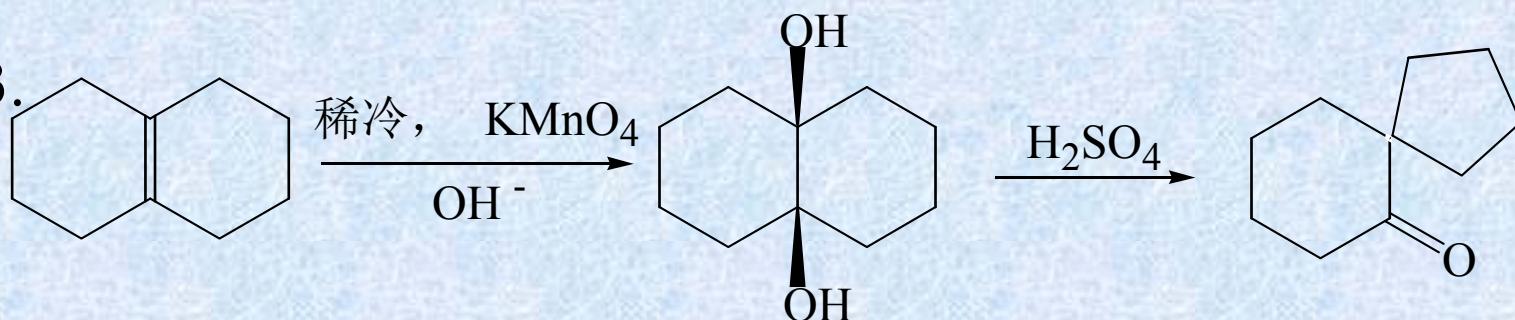


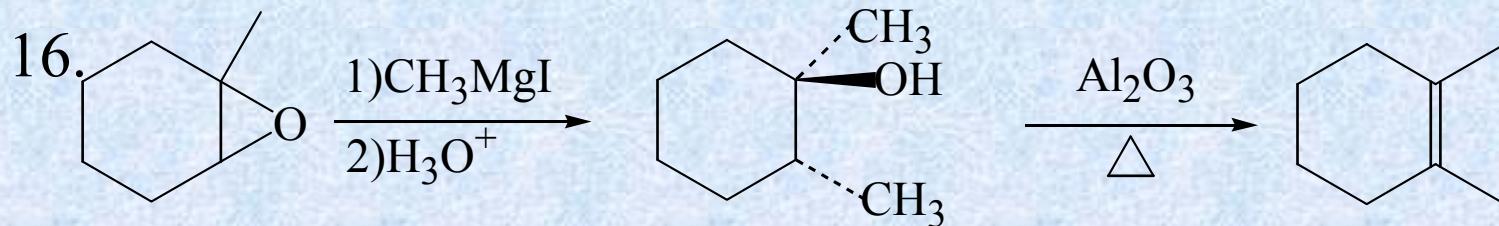
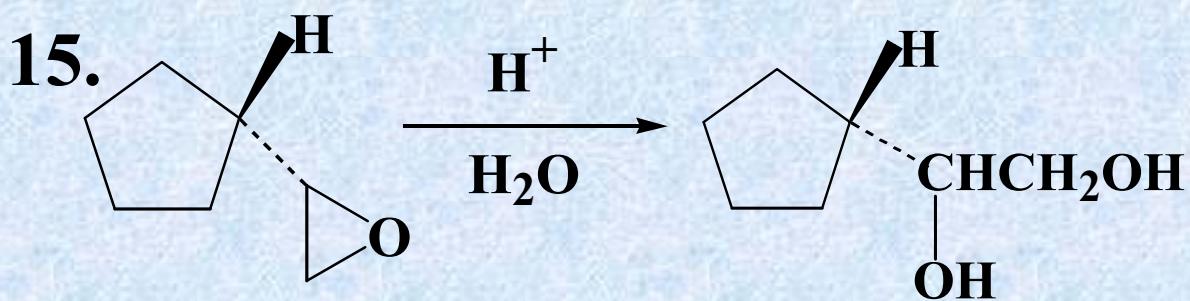
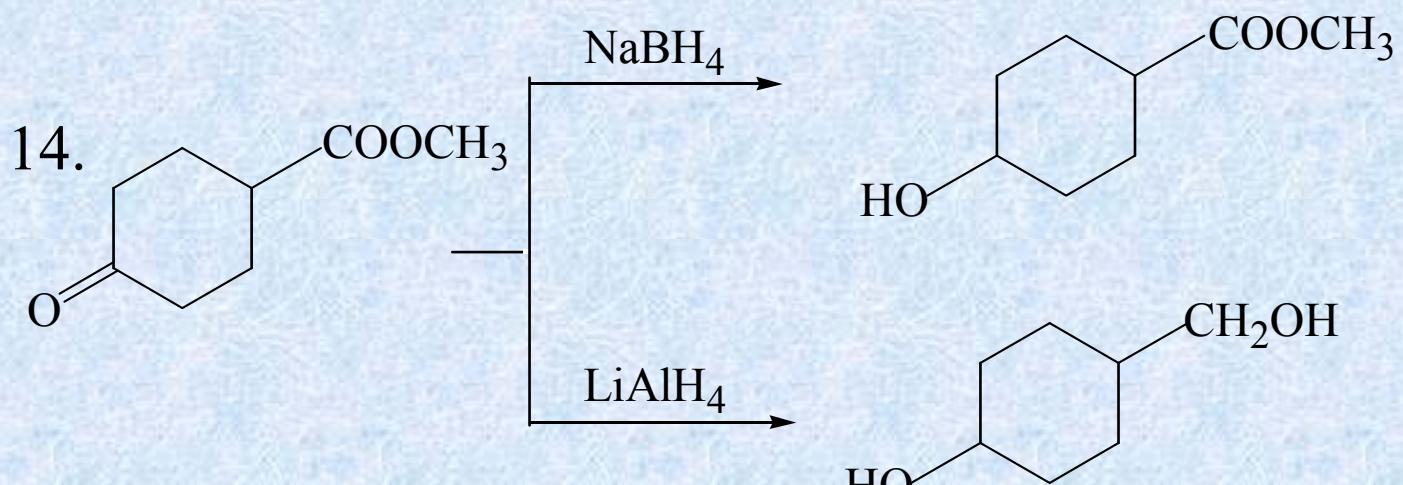


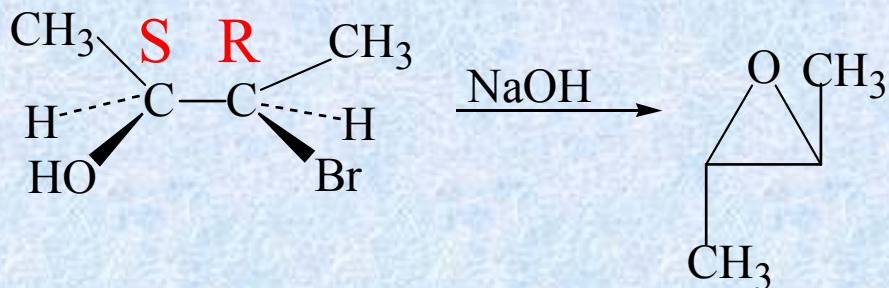
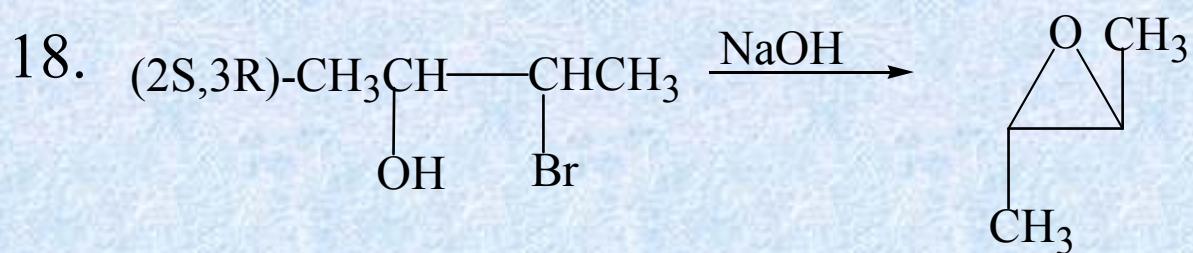
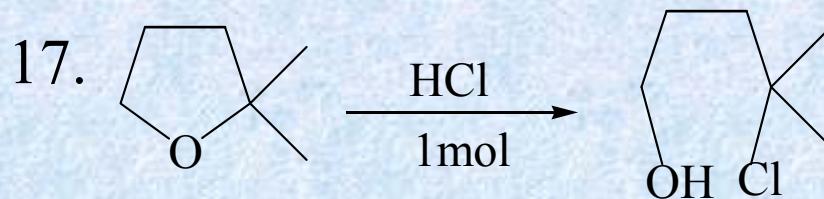




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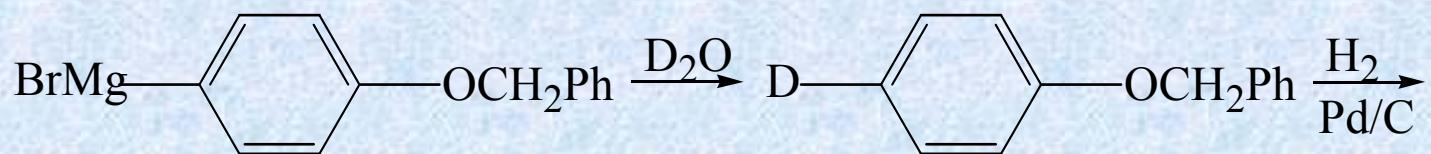
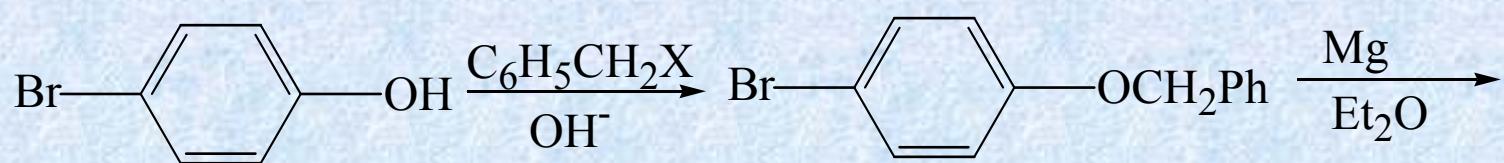
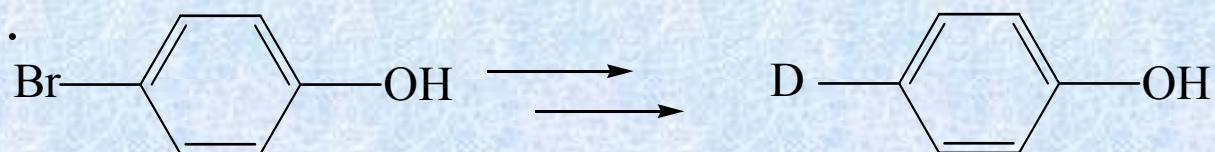




分子内的  $S_N2$  反应

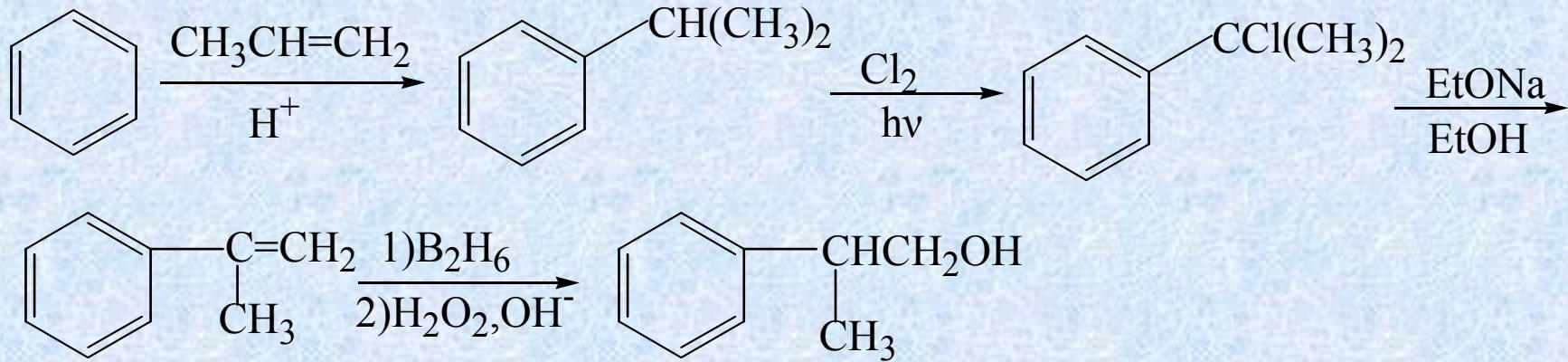
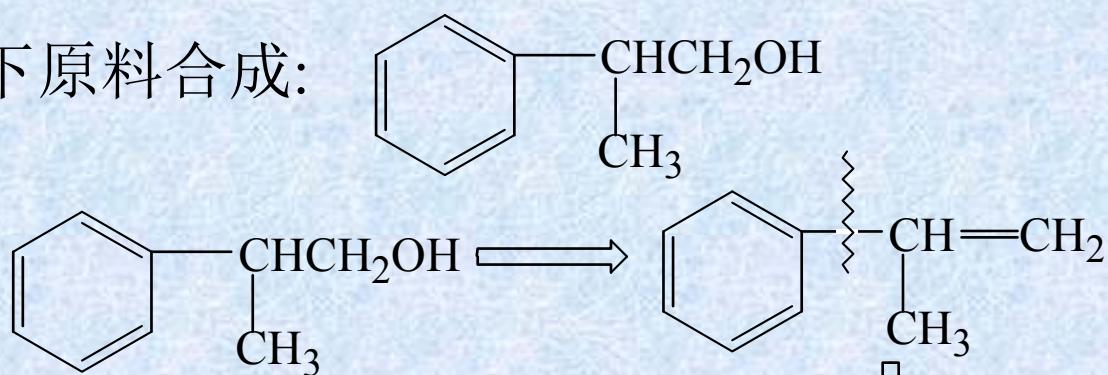
### 三. 合成:

1.

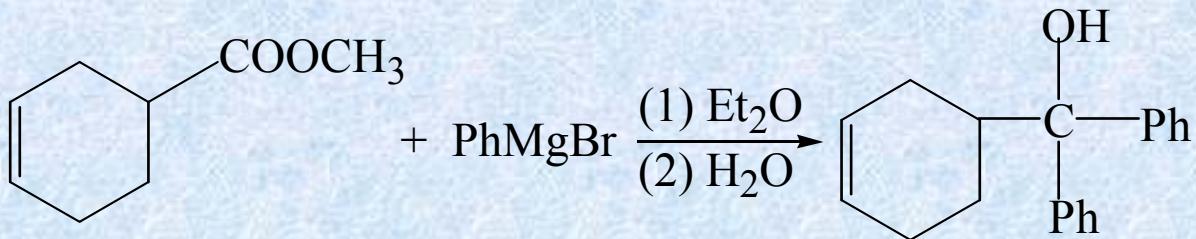
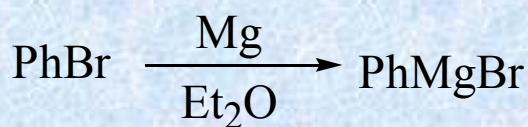
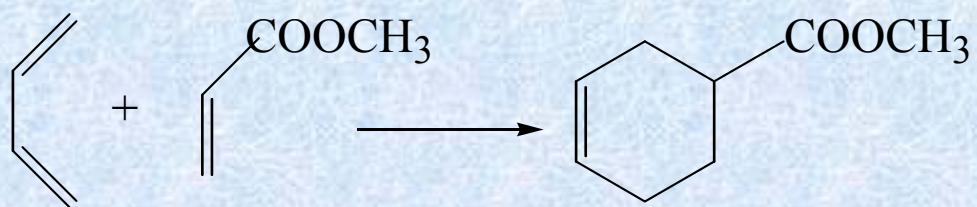
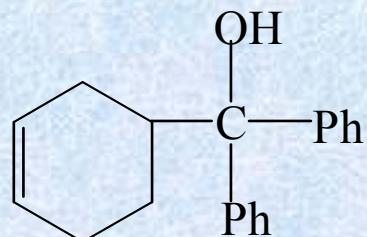


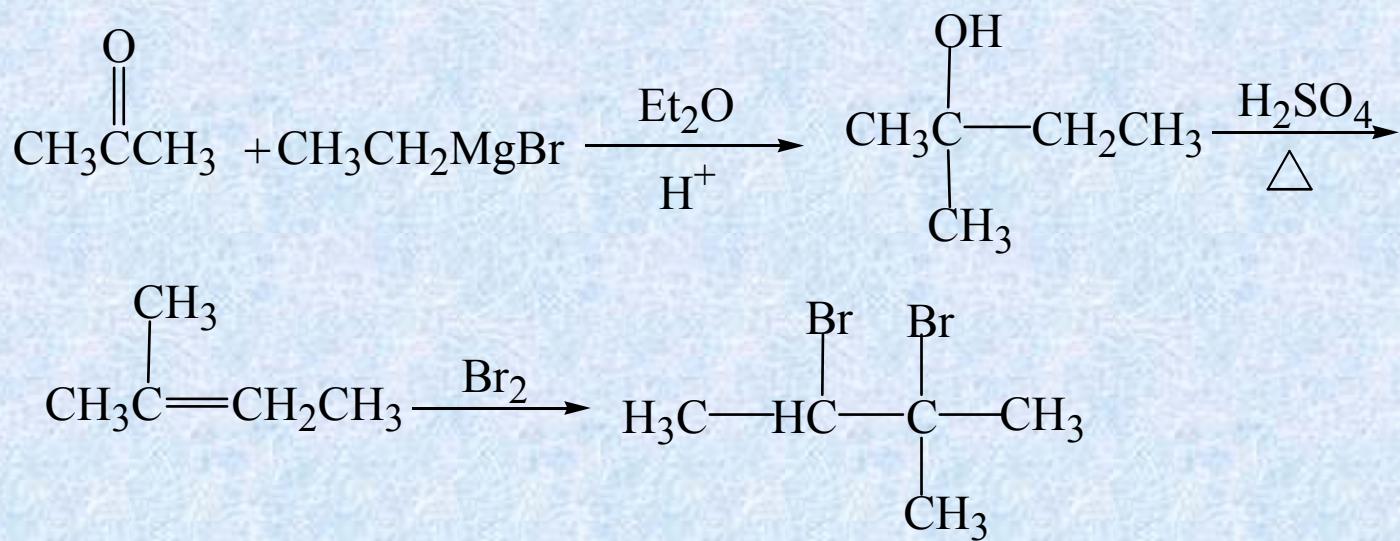
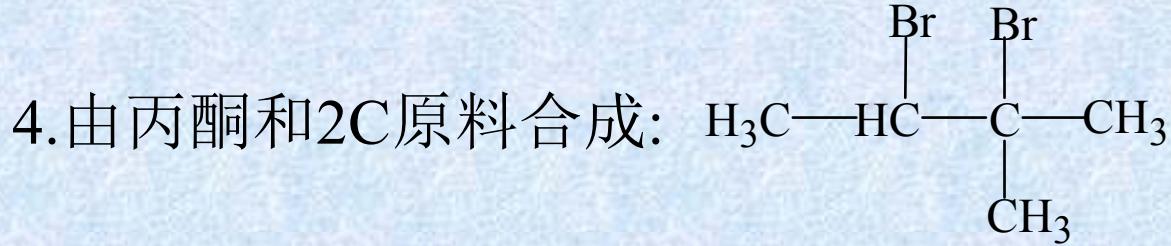
2.由苯和3C以下原料合成:

合成子分析:

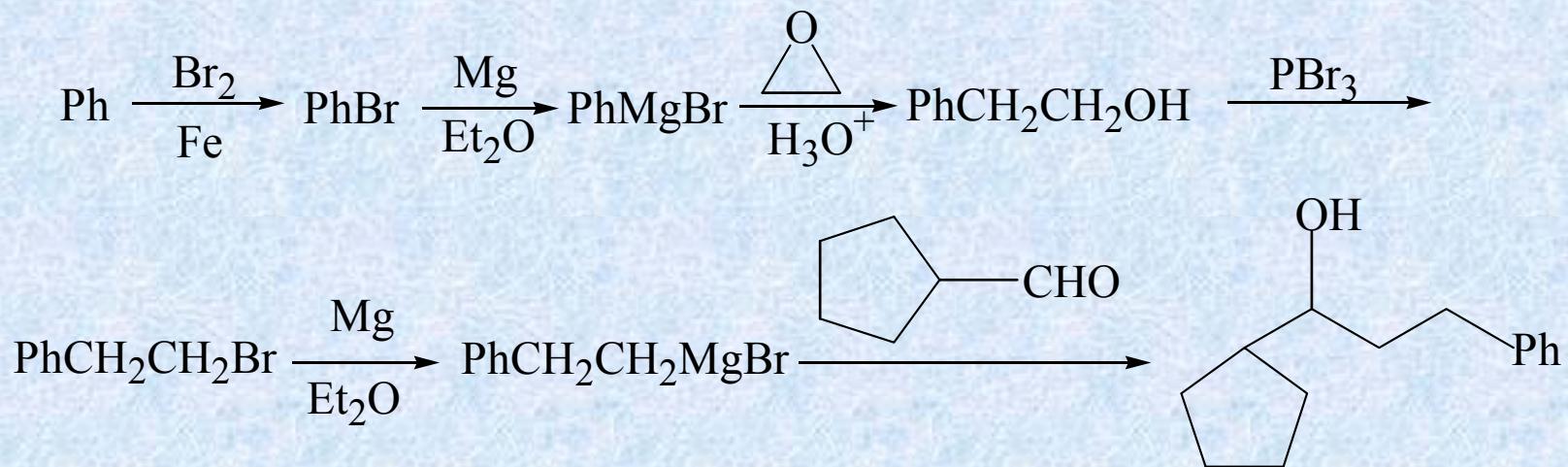
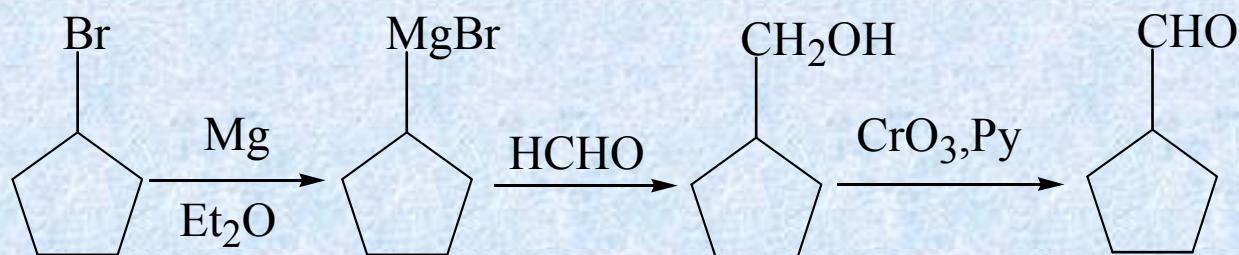
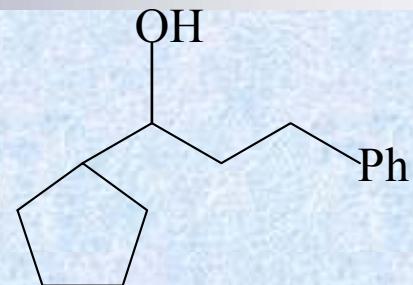


3.由C<sub>6</sub>H<sub>5</sub>Br 及4C以下原料合成:

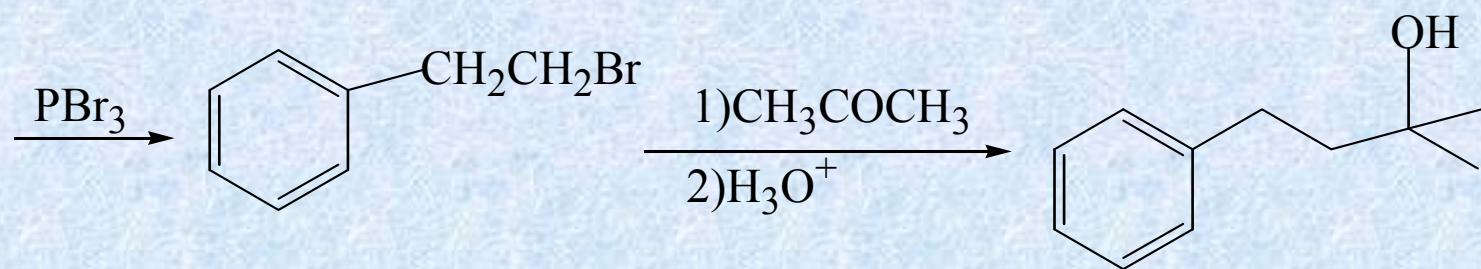
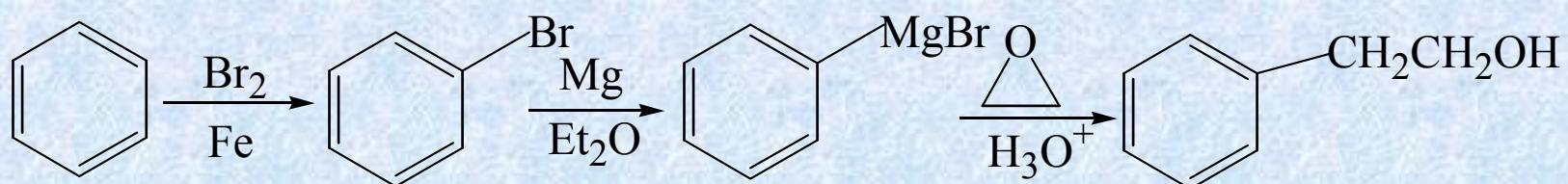
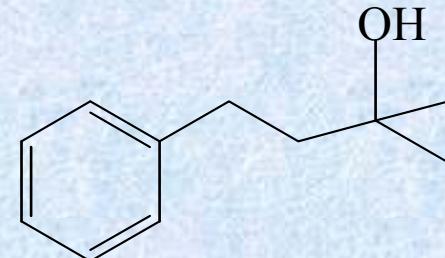


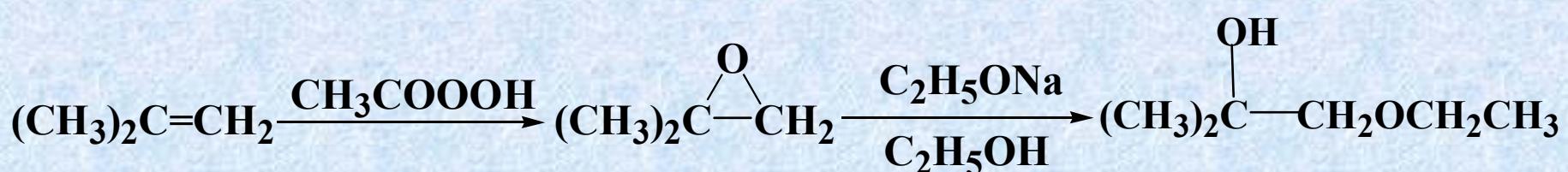
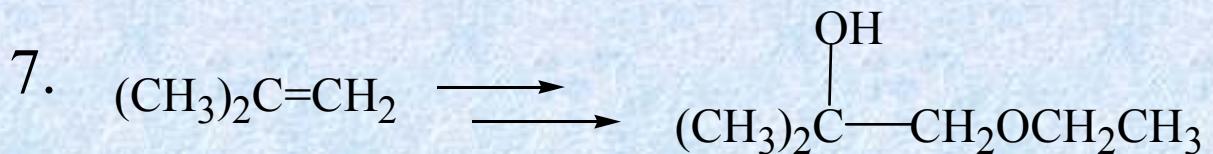


## 5.由苯及5C以下有机原料合成



6. 从苯, 环氧乙烷和丙酮合成:

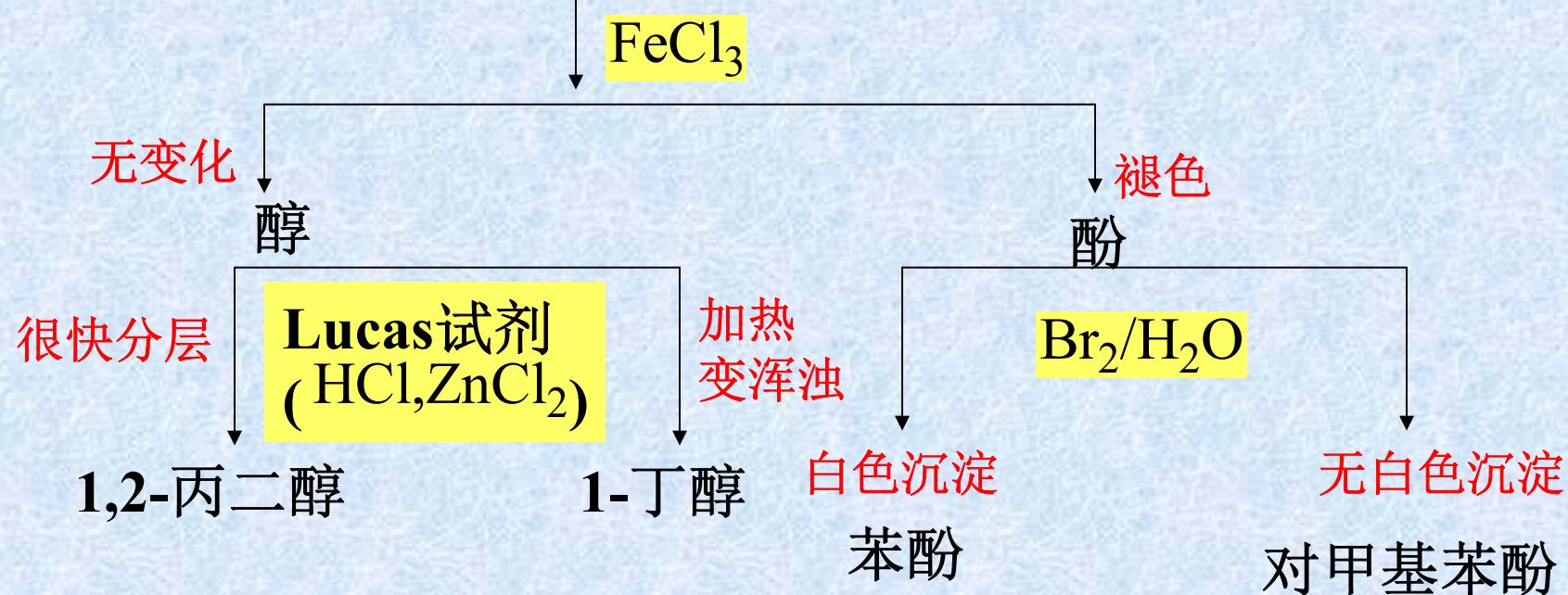




#### 四.用简单的化学方法区别下列4个化合物:

1,2-丙二醇、1-丁醇、苯酚、对甲基苯酚

1,2-丙二醇、1-丁醇、苯酚、对甲基苯酚



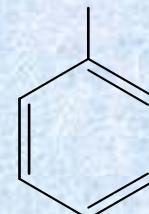
## 五、推理题：

1.  $C_{10}H_{14}O$ ,  $\nu_{max}$ : 3350, 1600, 490, 710, 690  $cm^{-1}$ ,  $\delta_H$ : 1.1(s, 6H), 1.4(s, 1H), 2.7(s, 2H), 7.2(s, 5H) ppm, 推测该化合物的结构。

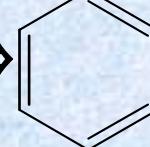
分析:  $C_{10}H_{14}O \Rightarrow$  不饱和度 = 4

IR:  $\nu_{max}$ : 3350  $cm^{-1} \Rightarrow -OH$

1600, 1490, 710, 690  $cm^{-1} \Rightarrow$



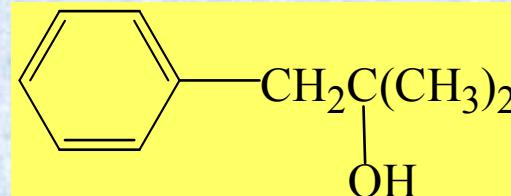
NMR:  $\delta_H$ : 7.2(s, 5H) ppm  $\Rightarrow$



2.7(s, 2H)  $\Rightarrow CH_2$

1.1(s, 6H)  $\Rightarrow 2CH_3$

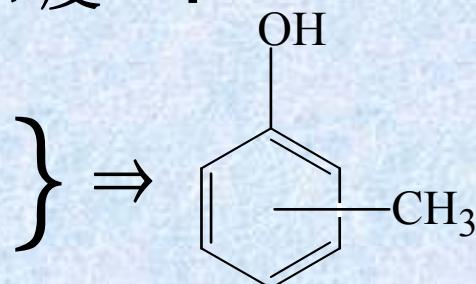
$\therefore$  此化合物为



2. 某化合物A, 分子式为 $C_7H_8O$ , 不溶于 $NaHCO_3$ 溶液, 但溶于 $NaOH$ 溶液, 当用溴水处理时, 迅速得到沉淀B, 分子式为 $C_7H_5OBr_3$ , 推出A, B的结构。

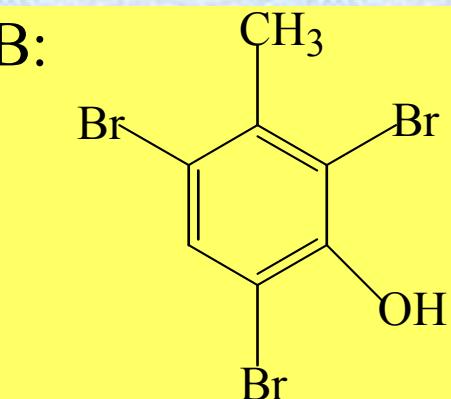
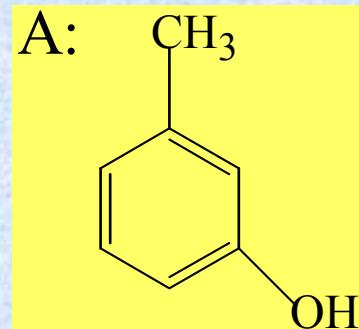
分析:  $C_7H_8O \Rightarrow$  不饱和度 = 4

不溶于 $NaHCO_3$ 溶液  
但溶于 $NaOH$ 溶液



A用溴水处理→沉淀B(分子式为 $C_7H_5OBr_3$ ) $\Rightarrow$  B:

$\therefore$



六. 写出下列反应的机理:

